LC-MS/MS Tandem Mass Spectrometry for Analysis of Phenolic Compounds and Pentacyclic Triterpenes in Antifungal Extracts of Terminalia brownii (Fresen)

Salih, Enass Yousif Abdelkarium

2017-12


http://hdl.handle.net/10138/232061
https://doi.org/10.3390/antibiotics6040037

Downloaded from Helda, University of Helsinki institutional repository.

This is an electronic reprint of the original article.

This reprint may differ from the original in pagination and typographic detail.

Please cite the original version.

Article

LC-MS/MS Tandem Mass Spectrometry for Analysis of Phenolic Compounds and Pentacyclic Triterpenes in Antifungal Extracts of *Terminalia brownii* (Fresen)

Enass Y. A. Salih 1,3,4,*, Pia Fyhrquist 3, Ashraf M. A. Abdalla 1, Abdelazim Y. Abdelgadir 1, Markku Kanninen 4, Marketta Sipi 4, Olavi Luukkanen 4, Mustafa K. M. Fahmi 1,4, Mai H. Elamin 5 and Hiba A. Ali 2,†

1 Department of Forest Products and Industries, Faculty of Forestry, PO Box 13314, University of Khartoum, Khartoum 11111, Sudan; amahmed@uofk.edu (A.M.A.A.); ayabdelgadir@uofk.edu (A.Y.A.); m kfahmi@uofk.edu (M.K.M.F.)
2 Commission for Biotechnology and Genetic Engineering, PO Box 2404, National Centre for Research, Khartoum, Sudan; hibaalihotmail.com
3 Faculty of Pharmacy, Division of Pharmaceutical Biosciences, PO Box 56, University of Helsinki, FIN-00014 Helsinki, Finland; pia.fyhrquist@helsinki.fi
4 Viikki Tropical Resources Institute (VITRI), Department of Forest Sciences, PO Box 27, University of Helsinki, FIN-00014 Helsinki, Finland; markku.kanninen@helsinki.fi (M.K.); Marketta.Sipi@helsinki.fi (M.S.); olavi.luukkanen@helsinki.fi (O.L.)
5 Department of Phytochemistry, Faculty of Pharmacy, PO Box 477, University of Sciences and Technology, Omdurman, Sudan; maielamin15@gmail.com
* Correspondence: enass.salih@helsinki.fi or eyabdelkareem@uofk.edu; Tel.: +358-46-9356-095
† With this paper we would like to honor our colleague, Dr. Hiba Ali, who passed away the 29.4.2016.

Academic Editor: Leonard Amaral
Received: 30 October 2017; Accepted: 7 December 2017; Published: 13 December 2017

**Abstract:** Decoctions and macerations of the stem bark and wood of *Terminalia brownii* Fresen. are used in traditional medicine for fungal infections and as fungicides on field crops and in traditional granaries in Sudan. In addition, *T. brownii* water extracts are commonly used as sprays for protecting wooden houses and furniture. Therefore, using agar disc diffusion and macrodilution methods, eight extracts of various polarities from the stem wood and bark were screened for their growth-inhibitory effects against filamentous fungi commonly causing fruit, vegetable, grain and wood decay, as well as infections in the immunocompromised host. Ethyl acetate extracts of the stem wood and bark gave the best antifungal activities, with MIC values of 250 μg/mL against *Nattrassia mangiferae* and *Fusarium verticillioides*, and 500 μg/mL against *Aspergillus niger* and *Aspergillus flavus*. Aqueous extracts gave almost as potent effects as the ethyl acetate extracts against the *Aspergillus* and *Fusarium* strains, and were slightly more active than ethyl acetate extracts against *Nattrassia mangiferae*. Thin layer chromatography, RP-HPLC-DAD and tandem mass spectrometry (LC-MS/MS), were employed to identify the chemical constituents in the ethyl acetate fractions of the stem bark and wood. The stem bark and wood were found to have a similar qualitative composition of polyphenols and triterpenoids, but differed quantitatively from each other. The stilbene derivatives, cis- (3) and trans- resveratrol-3-O-β-galloylglucoside (4), were identified for the first time in *T. brownii*. Moreover, methyl-(S)-flavogallonate (5), quercetin-7-β-O-di-glucoside (8), quercetin-7-O-galloyl-glucoside (10), naringenin-4′-methoxy-7-pyranoside (7), 5,6-dihydroxy-3′,4′,7-tri-methoxy flavone (12), gallagic acid dilactone (terminalin) (6), a corilagin derivative (9) and two oleanane type triterpenoids (1) and (2) were characterized. The flavonoids, a corilagin derivative and terminalin, have not been identified before in *T. brownii*. We reported earlier on the occurrence of methyl-S-flavogallonate and its isomer in the roots of *T. brownii*, but this is the first report on their occurrence in the stem wood as well. Our results justify the traditional uses of macerations and decoctions of *T. brownii* stem wood and
bark for crop and wood protection and demonstrate that standardized extracts could have uses for the eco-friendly control of plant pathogenic fungi in African agroforestry systems. Likewise, our results justify the traditional uses of these preparations for the treatment of skin infections caused by filamentous fungi.

**Keywords:** Africa; *Terminalia brownii*; antifungal stem wood and bark extracts; *Aspergillus; Nattrassia*; Fusarium; LC-MS/MS; flavonoids; ellagitannins; stilbenes; triterpenes

1. Introduction

Fungal contamination is both a pre- and a post-harvesting problem in crop production and poses a continuous and growing threat to global food crop production [1,2]. Some of the fungal species generally considered to be phytopathogens, such as *Aspergillus* spp., are also known to be increasingly significant as human pathogens, especially in the immunocompromised host [3–5].

*Aspergillus niger* (van Tieghem, 1867) and *Aspergillus flavus* (Link, 1809) are both human [6,7] and plant pathogens [8]. As human pathogens, especially *A. flavus*, but also *A. niger* cause aspergillosis in immunocompromised individuals [9,10]. *A. flavus* causes grain crop infections in maize (*Zea mays* L.), leading to a substantial decrease in the commercial value of maize crop due to aflatoxin contamination [11]. Moreover, *A. flavus* is often the causative agent of wood decay in timber and houses [12]. *Nattrassia mangiferae* [(Syd. and P. Syd.) B. Sutton and Dyko], previously known as *Hendersonula torulo* Nattrass (HT) and *Dothiorella mangiferae* (Syd. and P. Syd), is a wound-invading dematiaceous (brown-pigmented) phytopathogenic fungus infecting hard wood species of *Citrus*, *Mangifera* and *Eucalyptus* and soft wood coniferous subtropical and tropical trees, causing dieback and vascular wilt diseases [13,14]. *Nattrassia mangiferae* is also a human pathogenic fungus, especially in immunocompromised individuals [15], and is even known to cause community acquired infections in rural farmer societies worldwide [13]. *Fusarium verticilloides* and some other *Fusarium* species infect maize ears (husks) causing maize ear rot disease and contaminate maize grains with fumonisin mycotoxins leading to major pre- and post-harvest losses [16]. *Fusarium* spp. mycotoxins are toxic [17,18], and fumonisin has been found to cause cancer in mammalians [19]. Another species of *Fusarium*, *F. oxysporum* is the causative agent of the “Panama disease” affecting the banana (*Musa paradisiaca*), the staple food of a large part of Africa.

Currently used fungicides are costly and toxic to the environment [20,21]. Besides, phytopathogenic and human pathogenic filamentous fungi have developed resistance to many conventional fungicides and to antibiotics [22–26]. Thus, new effective, less toxic, affordable and readily available antifungals are needed [26]. Tropical and subtropical plants are known to contain a wide range of defense compounds due to their needs for constant production of defense compounds throughout the year as well as due to the high biodiversity in rain forests, woodlands and savannahs [27]. Thus, tropical plant species used for fungal infections in African traditional medicine as well as for protection of crop plants against fungal contamination, are expected to be good sources for new antifungal compounds [28,29].

The pantropical genus of *Terminalia* (Combretaceae) contains a number of species known for their antifungal effects. Antifungal activities against *Aspergillus niger* and *A. flavus* have been reported for the Asian species, *Terminalia alata*, *T. arjuna*, *T. bellerica*, *T. catappa* and *T. chebula* [30,31]. Approximately 30 species of *Terminalia* occur in Africa [32]. However, despite of their frequent uses in traditional medicine for treatment of fungal infections in humans and in traditional agriculture for prevention of fungal crop plant contamination, only a small portion of these species have been studied in depth for their antifungal activity and/or antifungal compounds. Among those African *Terminalia* species investigated for their antifungal potential, either against yeasts or filamentous fungi or both, are *T. avicennoides*, *T. spinosa*, *T. sericea* and *T. nigrovenulosa* [33–36]. Ellagitannins, ellagic acid derivatives, stilbenes, lignans, flavonoids and pentacyclic triterpenes were reported from some
African species of *Terminalia*, such as *T. horrida*, *T. sericea*, *T. superba* and *T. macroptera* [33,34,37–45]. Most of these phytochemical investigations did not include antifungal screening of the characterized compounds, however.

*Terminalia brownii* (Fresen.) is a deciduous tree distributed throughout East African savannah regions in a wide range of temperature, rain fall and soil conditions (Figure 1) [46]. In Sudan, *T. brownii* occurs in low and high rainfall zones in Blue Nile state and El-Gadarif in the eastern part of the country and Kordofan and Darfour states in the western part of Sudan. In Sudan, *T. brownii* grows in natural forested areas such as savannah woodlands where it is listed as an endangered species due to overexploitation [47]. *T. brownii* has been found to be exceptionally resistant against various pathogenic fungi that affect crops, and is frequently used in traditional agroforestry for crop plant protection [21]. Similarly to many other plant species, such as *Melianthus comosus* [48], also decoctions of various parts of *T. brownii* have traditional applications as fungicides against fungal contamination in harvested crop plants.

![Figure 1. Terminalia brownii. (A) tree in savannah woodland; (B) stem bark; (C) flowers and leaves; (D) fruits. Photo: E. Y. A. Salih and Dr. H. H. Gibreel, 2006.](image-url)

*Terminalia brownii* has been found to be a rich source of oleanane- and ursane-type pentacyclic triterpenoids, such as arjunic acid, galloyl arjunic acid, tomentosic acid, sericic acid, arjungenin, sericoside, betulinic acid, monogynol A and arjunglucoside [21,49]. In addition, a new oleanane type triterpenoid, designated as 3β,24-O-ethyldienyl-2α,19α-dihydroxoylean-12-en-28-oic acid, was identified in an ethyl acetate extract of the stem bark of *T. brownii* [49]. Moreover, β-sitosterol and ellagic acid derivatives have been characterized from the leaves and stem bark of *T. brownii* [21,49]. In addition, a number of unknown and known ellagitannins, including methyl-(S)-flavogallonate and its derivative as well as α,β-punicalagin and α,β-terchebulin have been described from the roots and the stem bark, respectively [50,51]. Moreover, a chromone derivative designated as terminalianone has been found in the stem bark of *T. brownii* [52]. However, to our knowledge, the antifungal effects of the mentioned compounds were not investigated, with the exception of arjungenin, β-sitosterol and betulinic acid [21].

Although *T. brownii* extracts are used traditionally against fungal phytopathogens and to treat human fungal infections in Sudan and in other countries of Africa, there are a limited number of reports on their in vitro antifungal activity against filamentous fungi affecting crop production and human health. Moreover, only a small number of antifungal compounds in *T. brownii* have been characterized to date [21,49], and to the best of our knowledge, no flavonoid structures have been studied in this species of *Terminalia*. Therefore, the current study was performed to verify the antifungal effects of decoctions and macerations, reported to be used for fungal infections in traditional medicine and as fungicides in traditional agriculture. In addition, extracts of various polarities made from the stem bark and wood of *T. brownii* were tested for their growth-inhibitory effects. For the screenings, significant phytopathogenic and human pathogenic opportunistic fungi of the genera *Aspergillus*, *Nattrassia* and...
2. Results

2.1. Antifungal Effects of Extracts of Terminalia brownii Stem Bark and Wood

The results of the growth inhibition of various extracts of Terminalia brownii stem bark and wood against Aspergillus, Nattrassia and Fusarium strains are shown in Table 1. When compared to the other extracts, the ethyl acetate extracts of the stem wood and bark gave the highest antifungal activity. This result is in accordance with other authors, who also reported that especially ethyl acetate extracts of the stem bark of T. brownii give good antifungal effects against sweet potato infecting fungi, such as Aspergillus niger and Fusarium solani [21]. The reference antifungal used in our tests, amphotericin-B, was more growth-inhibitory than the ethyl acetate extracts, however, although the differences in potency between amphotericin-B and the ethyl acetate extracts were not big, considering that we used plant extracts instead of pure compounds present in these extracts (Table 1).

Our results demonstrate that Nattrassia mangiferae and Fusarium verticilloides were especially sensitive to the ethyl acetate extracts of T. brownii stem wood and bark, giving MIC values of 250 µg/mL, whereas Aspergillus niger and A. flavus were more resistant to these extracts, demonstrating MIC values of 500 µg/mL (Table 1). We found that the obtained MIC values correlated well with the sizes of the inhibition zones produced by these ethyl acetate extracts, so that small MIC values were coupled to large diameters of the inhibition zones (Table 1). To the best of our knowledge, this is the first report on antifungal effects of T. brownii against Nattrassia mangiferae.

Interestingly, we also found that aqueous extracts of the stem bark and wood of T. brownii gave good growth-inhibitory effects (Table 1). Compared to the other extracts these aqueous extracts gave especially high extraction yields of 20 and 16%, respectively for bark and wood (Figure 2). Thus, our results justify the traditional application of macerations (water extracts) of the stem wood and barks of T. brownii for the preservation of grains and wooden house poles and for traditional medicinal treatment of fungal infections caused by Aspergillus, Nattrassia and Fusarium. Earlier studies have indicated that aqueous stem bark extracts of T. brownii are growth-inhibitory also against yeast species, such as Candida albicans and Cryococcus neoformans and in addition the aqueous extracts were found to be less toxic than other extracts against brine shrimps [53]. Thus, standardized aqueous extracts of T. brownii stem wood and bark could be used to treat fungal infections and fungal contamination.

Figure 2. Percentage yield (% w/w) resulting from sequential extraction and liquid/liquid partition of the stem wood and stem bark of Terminalia brownii. W, stem wood; B, stem bark; hex, hexane extract; Ch, chloroform extract; EtOAc, ethyl acetate extract; aqu, aqueous extract.
When compared to the more polar water and ethyl acetate extracts, we found that chloroform extracts of the stem wood and bark of *T. brownii* were slightly antifungal, while the petroleum ether extracts were devoid of antifungal activity (Table 1). In contrast to our results, in an earlier investigation, it was found that an n-hexane extract of the stem bark of *Terminalia brownii* was active against another *Aspergillus* species, *A. fumigatus* [54]. Perhaps this result might indicate that different species of *Aspergillus* differ to their sensitivity to non-polar extracts of *T. brownii*, so that *A. fumigatus* is more sensitive than *A. niger* and *A. flavus*.

Table 1. Antifungal activity of stem wood and bark extracts of *T. brownii*. Results were obtained using cup well agar diffusion and agar dilution methods.

<table>
<thead>
<tr>
<th>Fungal Strain</th>
<th>Stem Wood Extracts</th>
<th>Stem Bark Extracts</th>
<th>Amphotericin-B</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>IZ</td>
<td>MIC</td>
<td>IZ</td>
</tr>
<tr>
<td><em>Aspergillus niger</em></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Pt CHCl3</td>
<td>NA</td>
<td></td>
<td>NA</td>
</tr>
<tr>
<td>EtOAc</td>
<td>17 ± 0.7</td>
<td>500</td>
<td>17 ± 0.8</td>
</tr>
<tr>
<td>aqueous</td>
<td>17 ± 0.5</td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Aspergillus flavus</em></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Pt CHCl3</td>
<td>14 ± 0.5</td>
<td></td>
<td>14 ± 0.9</td>
</tr>
<tr>
<td>EtOAc</td>
<td>18.5 ± 0.4</td>
<td>500</td>
<td>18.5 ± 0.8</td>
</tr>
<tr>
<td>aqueous</td>
<td>18 ± 0.9</td>
<td></td>
<td>18 ± 0.5</td>
</tr>
<tr>
<td><em>Nattrassia mangifea</em></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Pt CHCl3</td>
<td>12 ± 0.5</td>
<td></td>
<td>12 ± 0.7</td>
</tr>
<tr>
<td>EtOAc</td>
<td>19 ± 0.4</td>
<td>250</td>
<td>18.5 ± 0.4</td>
</tr>
<tr>
<td>aqueous</td>
<td>18.5 ± 0.4</td>
<td></td>
<td></td>
</tr>
<tr>
<td><em>Fusarium verticillioides</em></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Pt CHCl3</td>
<td>13 ± 0.6</td>
<td></td>
<td>11 ± 0.9</td>
</tr>
<tr>
<td>EtOAc</td>
<td>20 ± 0.4</td>
<td>250</td>
<td>19 ± 0.2</td>
</tr>
<tr>
<td>aqueous</td>
<td>19 ± 0.3</td>
<td></td>
<td>18 ± 0.7</td>
</tr>
</tbody>
</table>

For agar diffusion, extracts at the concentration 1 mg/mL were used. Diameter of inhibition zones (IZ) in mm: >18 mm: sensitive; 14–18 mm: intermediate; <14 mm: resistant [55,56]; Pt, petroleum ether extracts; CHCl3, chloroform extracts; EtOAc, ethyl acetate extracts; NA, Not active. IZ results as mean ± SEM of five measurements. MIC in µg/mL. The observed differences between the sample means of the inhibition zones (the stem bark and wood extracts) against the tested fungi did not differ significantly.

2.2. Results from the Phytochemical Screening of Antifungal Ethyl Acetate Extracts of *T. brownii* Stem Wood and Bark

Owing to our promising antifungal results for the ethyl acetate extracts of the stem bark and wood of *T. brownii*, and to the few existing earlier records on the activity of this species against filamentous fungi, we investigated the secondary compound composition and molecular masses as well as the fragmentation patterns of phenolic compounds and triterpenoids of these extracts.

2.2.1. TLC Results

RP-18 thin layer chromatograms of the ethyl acetate extracts of the stem wood and stem bark of *T. brownii* gave a negative reaction with Dragendorff reagent, suggesting that these extracts were devoid of alkaloids. Pink to purple colors were developed upon spraying with vanillin-H$_2$SO$_4$, which suggested the presence of triterpenoid and phenolic compounds. Spraying the TLC plates with aluminum trichloride (AlCl$_3$) and Natural Product reagent (NPR), revealed the presence of flavonoids, since color changes from quenching fluorescence to yellow, orange or blue color, typical for flavonoidal acids or other phenolic acids, could be observed at 366 nm [57].
2.2.2. HPLC-UV/DAD Results

HPLC-UV/DAD fingerprints of the ethyl acetate extracts of the stem bark and stem wood of *Terminalia brownii* are presented in Figure 3. Altogether twelve compounds with retention times between 6.8 and 25.5 min could be identified using internal standards and a computer library for standard compounds. At the wavelengths of 320 and 254 nm, which were used for detection of stilbenes and flavonoids, the wood ethyl acetate extract displayed a higher diversity of flavonoidal and stilbenoid compounds. For example, the cis- and trans-isomers of resveratrol 3-O-β-galloyl-glucoside (3 and 4) at Rt 11.1 and 13.2 min, respectively, as well as naringenin-4′-methoxy-7-pyranoside (7) at 15.3 min, the corilagin derivative (9) at Rt 18.2 min, and quercetin-7-O-galloyl glucoside (10) at Rt 18.4 min, were present in the wood extract but absent from the stem bark extract as shown in Figure 3.

![Figure 3. RP-HPLC/DAD chromatograms of ethyl acetate extracts of *T. brownii*. (A) stem bark and (B) stem wood extracts at 254 nm. (1) and (2) Oleanane type triterpenoids; (3) cis-resveratrol-3-O-β-galloyl-glucoside; (4) trans-resveratrol-3-O-β-galloyl-glucoside; (5) Methyl-(S)-flavogallonate; (6) Gallagic acid dilactone (Terminalin); (7) Naringenin-4′-methoxy-7-pyranoside; (8) Quercetin-7-β-O-diglucoside; (9) Corilagin derivative; (10) Quercetin-7-O-galloyl-glucoside; (11) unknown ellagitannin; (12) 5,6-dihydroxy-3′,4′,7-trimethoxy flavone.](image)

Because of the high number of compounds present in the wood ethyl acetate extract and due to this extract being slightly more antifungal than the stem bark, at least in terms of the sizes of the diameters of inhibition zones (Table 1), this extract was subjected to LC-MS/MS advanced analysis for identification of the major compounds.

2.2.3. LC-MS/MS Results

MS/MS combined with collision-induced dissociation (CID), has been found to enable the accurate identification of stilbenes and flavonoids in complex extracts with co-eluting peaks [58]. Therefore MS/MS was employed as the method of choice for the identification of compounds in an ethyl acetate extract of *T. brownii* stem wood. A total of twelve compounds were characterized by comparing the obtained molecular (precursor) ions and fragmentation patterns (i.e., product ions) from our LC-MS/MS data with data from the literature and with a computer library for the standard compounds.
(Table 2). The MS², MS³ and MS⁴ ion chromatograms are presented in the supplemental part of this paper (Supplement 1).

**Table 2.** HPLC-DAD and MS/MS data of phenolic compounds and triterpenoids in an ethyl acetate extract of the stem wood of *T. brownii*.

<table>
<thead>
<tr>
<th>Peak No</th>
<th>Rt (min)</th>
<th>[M-H]⁻ (m/z)</th>
<th>CID Mⁿ Main Fragment Ions (m/z)</th>
<th>Identified Compound</th>
<th>Molecular Formula</th>
<th>Exact Mass (Calc.)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>6.8</td>
<td>469</td>
<td>425, 407, 379, 355, 300, 271</td>
<td>oleanane type triterpenoid</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>2</td>
<td>6.8</td>
<td>491</td>
<td>447, 429, 411, 401, 385, 301</td>
<td>oleanane type triterpenoid</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>3</td>
<td>11.1</td>
<td>541</td>
<td>532, 425, 397, 301, 273, 227, 199, 169</td>
<td>cis-resveratrol-3-O-β-galloyl-glucoside</td>
<td>C₂₂H₂₆O₁₂</td>
<td>542.1416</td>
</tr>
<tr>
<td>4</td>
<td>13.2</td>
<td>541</td>
<td>532, 424, 407, 300, 275, 227, 199, 169</td>
<td>trans-resveratrol-3-O-β-galloyl-glucoside</td>
<td>C₂₂H₂₆O₁₂</td>
<td>542.1416</td>
</tr>
<tr>
<td>5</td>
<td>14.1</td>
<td>483</td>
<td>451, 433, 407, 305, 405, 377</td>
<td>cis-β-flavogallonate</td>
<td>C₁₉H₁₂O₆</td>
<td>542.1416</td>
</tr>
<tr>
<td>6</td>
<td>14.4</td>
<td>601</td>
<td>583, 301, 299, 271, 243, 215</td>
<td>Gallagic acid dilactone</td>
<td>C₂₈H₁₀O₆</td>
<td>601.9964</td>
</tr>
<tr>
<td>7</td>
<td>15.3</td>
<td>433</td>
<td>300, 314, 229, 271, 132</td>
<td>Naringenin-4′-methoxy-7-pyranoside</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>8</td>
<td>16.8</td>
<td>625</td>
<td>301, 284, 256, 229, 201,185, 129</td>
<td>Quercetin-7-β-O-diglucoside</td>
<td>C₂₀H₁₆O₁₇</td>
<td>626.1473</td>
</tr>
<tr>
<td>9</td>
<td>18.2</td>
<td>633</td>
<td>481, 463, 421, 387, 305, 275, 300, 169</td>
<td>Corilagin derivative</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>10</td>
<td>18.4</td>
<td>585</td>
<td>301, 284, 257, 229, 201, 185, 153, 132</td>
<td>Quercetin-7-O-galloyl-glucoside</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>11</td>
<td>19.1</td>
<td>725</td>
<td>665, 503, 409, 441, 379, 391</td>
<td>Unknown ellagitannin</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>12</td>
<td>25.5</td>
<td>343</td>
<td>328, 313, 298, 285, 270, 257</td>
<td>5,6-dihydroxy-3′,4′,7-trimethoxy-flavone</td>
<td>-</td>
<td>-</td>
</tr>
</tbody>
</table>

Rt, retention time in HPLC-DAD; [M-H]⁻ (m/z), base or molecular ions at negative mode; CID Mⁿ, Fragmentation ions resulting from collision-induced dissociation; The exact mass (calc.) according to the molecular formula of identified compounds. Aglycones are underlined. Peak numbers according to Figure 3.

We found that the stem wood of *T. brownii* contains two oleanane triterpenoid acids that co-eluted at 6.8 min (Figures 3 and 4). For compound (1) a [M-H]⁻ molecular ion at m/z 469 was detected, whereas compound (2) gave a molecular ion of m/z 491. In the MS² chromatograms, a fragment ion at m/z 425 was detected for compound (1) and at m/z 447 for compound (2) (Table 2, Supplement 1, Slide 1). These fragment ions indicate the loss of a carboxylic acid (−COOH) group ([M-H]⁻ for −COOH = 44) from both molecular ions. In agreement with our results, the loss of carboxylic acid at position 17 in pentacyclic triterpenoids was observed when using atmospheric pressure chemical ionization (APCI)-MS [59,60]. Moreover, we observed a fragment ion at m/z 407, which indicated the loss of H₂O from m/z 425 in compound 1. This kind of mass spectral fragmentation pattern is typical for oleanane type triterpenes [61,62], therefore confirming that compounds 1 and 2 are oleanane type triterpenes (Figure 4A).

In our HPLC-DAD system, compounds (3) and (4) eluted at Rt 11.1 and 13.2 min, respectively (Figure 3b). Both compounds showed an identical [M-H]⁻ molecular ion at m/z 541. Moreover, when subjected to MS³, both compounds provided fragment ions of m/z 227 and 314 (Table 2, Supplement 1, Slide 3 and 4). The fragment ion at m/z 314 indicates the presence of a galloylhexose fragment [63]. A comparison with the literature showed that the fragment ion at m/z 227 corresponds to the resveratrol unit [44]. Therefore, compounds (3) and (4) were tentatively assigned as resveratrol-3-O-β-galloyl-glucoside, respectively. Due to different retention times, the compounds were proposed to be *cis*- (3) and *trans*- (4) isomers of resveratrol-3-O-β-galloylglucoside (Figure 4C).
When subjected to MS², compound (5) at Rt 14.1 min in HPLC-DAD (Figure 3b), gave an [M-H]⁻ molecular ion at m/z 483 (Table 2, Supplement 1, Slide 5). The loss of the two oxygen molecules [483 − 451 = 32] at MS³, gave a fragmentation ion at m/z 433. Also, the MS³ spectrum was devoid of the fragment of [M-H-CO₂]⁻, which corresponds to a methyl ester molecule [64]. Therefore, and according to previous investigations [45,50], compound (5) was tentatively assigned the structure of methyl-(S)-flavogallonate.

Compound (6) at Rt 14.4 min (Figure 3b), gave a [M-H]⁻ molecular ion at m/z 601 (Figure 3b, Table 2). When subjected to MS³, compound (6) yielded the fragment ions at m/z 271 and 301, the later corresponding to free ellagic acid (Supplement 1, Slide 6). A molecular ion of m/z 601 and fragment ions at m/z 271 and 301 have been reported for gallagic acid [45]. Gallagic acid dilactone (syn. terminalin) has been reported in another species of *Terminalia, T. oblongata* [65]. Accordingly, compound (6) was tentatively assigned the structure of gallagic acid dilactone (terminalin) (Figure 4B).

Compound (7) at Rt 15.3 min (Figure 3b) gave an [M-H]⁻ molecular ion at m/z 433 (Table 2). The deprotonation of [M-H]⁻ at MS² resulted in a fragment ion at m/z 300, indicating the loss of one molecule of a pentose sugar ([M-H]-132) [63,66]. Moreover, MS³ of this compound yielded the loss of an Y0 fragment at m/z 271, corresponding to the cleavage of the aglycone fragment ion of the flavanone naringenin [66–68]. In the MS³, the [M-H]⁻ yielded a fragment at m/z 284, therefore indicating that the methoxy group occurs at position 4’ [66] (Supplement 1, Slide 7). As flavonoids commonly occur as O-glycosides and O-glycosylation occurs at position 7 in flavanones [63,69], compound (7) was tentatively assigned to naringenin-4’-methoxy-7-pyranoside.

The main compound (8) in the HPLC chromatogram at Rt 16.8 min (Figure 3) gave a [M-H]⁻ molecular ion at m/z 625 (Table 2). The main fragmentation product ion at m/z 301 in the MS² and MS³ chromatograms indicated the loss of two glucose molecules ([M-H]⁻-2 × 162 Da) as well the presence of a quercetin aglycone moiety corresponding to m/z 301 [68] (Supplement 1, Slide 8). Since glucose is usually β-glycosidically linked to the flavonoid aglycone and O-glycosidic linking is usually occurring at position 7 on the A ring of flavonoids [69], compound (8) was tentatively identified as quercetin-7-β-O-diglucoside.

Compound (9) at Rt 18.2 min (Figure 3b), gave a [M-H]⁻ molecular ion at m/z 633 (Table 2). MS/MS fragmentation resulted in a loss of a fragment product ion at m/z 481, corresponding to [M-galloyl-gallic acid. Other fragment product ions resulting from MS² were; 463 [M-gallic acid]⁻, 300 (hexahydroxydiphenolyl-H) and 169 corresponding to gallic acid [45,70,71] (Supplement 1, Slide 9). Consequently, compound (9) is suggested to be a derivative of corilagin.

Compound (10) at Rt 18.4 min (Figure 3b) gave a [M-H]⁻ molecular ion at m/z 585 (Figure 3, Table 2). MS² fragmentation of this compound resulted in the loss of a pyranose sugar corresponding to the fragment ion at m/z 132 and a galloyl unit corresponding to a fragment ion of m/z 153. Moreover, a fragment product ion at m/z 301 ([M-H]⁻-132-153) (Table 2), corresponding to the aglycone
of quercetin, was present in the MS² chromatogram [68] (Supplement 1, Slide 10). Consequently, compound (10) was tentatively assigned to be quercetin 7-O-galloyl-glucoside.

A polyphenol (Compound 11) with a retention time of 19.1 min in HPLC-DAD (Figure 3b) gave a molecular ion at m/z 725. In the MS² spectrum, the fragment ions of a hexose sugar were observed at m/z 665 and 503, indicating that the cleavage within this hexose sugar ring occurred at 0.3X [M-H-61]⁻ [63,69]. Moreover, the loss of two oxygen molecules was noticed at MS³ [M-H-2 × 16]⁻. In the spectra at MS⁴, two fragments at m/z 391 and 379 were observed. These fragments resulted from the loss of one molecule of water and two methyl groups, respectively, from the fragment at m/z 409 in spectra MS³ ([M-H]⁻-H₂O-2CH₃)⁻ (Supplement 1, Slide 11). Thus, compound (11) is suggested to be identical to an unknown ellagitannin that we have reported earlier to occur in T. brownii roots [50].

Compound (12) at Rt 25.5 min (Figure 3b), gave a [M-H]⁻ molecular ion at m/z 343 (Table 2). MS³ and MS⁴ fragmentation of this compound resulted in the loss of three methyl groups (-CH₃) corresponding to product fragment ions ([M-H]⁻-343-328-313) (Table 2, Supplement 1, Slide 12). Moreover, in the fragment ion chromatogram resulting from MS³, a high intensity of the product fragment ion at m/z 313 could be observed indicating the loss of a methyl group ([M-H]⁻-15) (Table 2, Supplement 1, Slide 12). From this data, compound (12) was tentatively assigned as 5,6-dihydroxy-3′,4′,7-trimethoxyflavone.

3. Discussion

Pentacyclic triterpene saponins are known to complex with ergosterol and cholesterol in the fungal cell membrane, thus leading to loss of membrane integrity [72] and it has been found that triterpenoids decrease mycelial growth [73]. Accordingly, the triterpenes betulinic acid and arjungenin, isolated from ethyl acetate extracts of T. brownii stem bark were found to give good antifungal effects against Aspergillus niger, Fusarium solani and Fusarium oxysporum with MIC values ranging from 50 to 200 µg/mL [21]. Therefore, we suggest that the two unknown oleanane-type triterpenes (1) and (2) would contribute significantly to the antifungal effects we have found for the ethyl acetate extract of the stem wood of T. brownii (Table 2, Figure 3).

In the genus Terminalia, resveratrol and its glucoside and rutinoside derivatives have been reported in Terminalia prunioides, T. sericea and T. ferdinandiana [44,74,75]. We reported here for the first time on the occurrence of the resveratrol derivatives, cis- (3) and trans-resveratrol-3-O-β-galloylglucoside (4) in Terminalia brownii stem wood. Besides, galloylglucoside derivatives of resveratrol have not been reported before in the genus Terminalia. Resveratrol and its derivatives are antifungal phytoalexins, protecting plants from pathogenic fungal and bacterial intrusion [76,77]. Several investigations on in vitro antifungal activities of resveratrol and its derivatives indicate good antifungal potential of this compound class [78,79]. Therefore, the good antifungal activity in the ethyl acetate extracts of T. brownii could partly be due to the resveratrol-galloylglucoside derivatives (2) and (3). To the best of our knowledge resveratrol-3-O-β-galloylglucoside has not been studied for its antifungal effects, which warrants further studies in this respect.

We reported here for the first time, on the occurrence of another ellagic acid derivative, gallic acid dilactone (6), in the stem wood of T. brownii. Gallic acid, is an analogue to ellagic acid, containing four gallic acid residues [80] and has restricted occurrence in plants. Gallic acid and its derivatives have been found in various parts of some other Terminalia species such as in the leaves of T. catappa and T. oblongata [65,81] and in the fruits of Terminalia bellerica, Terminalia chebula and T. chebula [45]. Gallic acid is the fully lactonized form of the gallagyl moiety in the ellagitannin punicalagin, which is common in Terminalia spp. [45,82]. Gallic acid has been found to give concentration-dependent growth-inhibitory effects against Fusarium and Alternaria [83]. Thus, it is possible that gallic acid dilactone (6), which we found to be present in an ethyl acetate extract of the stem wood of T. brownii, could be an important contributor to the antifungal effects of this extract.

Ellagitannins have been found to inhibit the growth of Fusarium and Alternaria dose-dependently [83]. Even though the genus Terminalia is renowned to be especially rich in
ellagitannins [45] only a few studies have been performed on the antifungal effects of ellagitannins isolated from Terminalia species. Some of the few investigations demonstrate that ellagitannins from Terminalia spp. could be valuable antifungal compounds. For example, punicalagin from the leaf of T. brachystemma was found to give a low MIC value of 6.25 µg/mL against Candida strains [61]. It was found, however, that some ellagitannins were not active against filamentous fungi, although activity was demonstrated against Candida and Cryptococcus neoformans [84]. Thus, it remains to be investigated whether the ellagitannins we have found in T. brownii stem wood, such as methyl-(S)-flavogallonate (5), the unknown ellagitannin (11) and the corilagin derivative (9) give low MIC values against filamentous fungi such as Aspergillus, Nattrassia and Fusarium spp. among others.

Our research resulted in the characterization of the flavonoids naringenin-4′-methoxy-7-pyranoside (7), quercetin-7-β-O-diglucoside (8), quercetin-7-O-galloylglycoside (10) and 5,6-dihydroxy-3,4,7-trimethoxy flavone (12) in the ethyl acetate extracts of stem wood and bark of Terminalia brownii. To the best of our knowledge this is the first time these flavonoids are reported to occur in T. brownii. We suggest that quercetin-7-β-O-diglucoside (8), which was quantitatively the main peak in both stem bark and wood extracts of T. brownii, contributes significantly to the antifungal effects of these extracts. Accordingly, several authors have reported that quercetin and its derivatives give good antifungal effects against Aspergillus and Fusarium strains [85–87] and for quercetin as low MIC values as 15 µg/mL were recorded against Aspergillus niger, Fusarium moniliforme and F. sporotrichum [88]. Furthermore, dihydroquercetin from barley suppressed the growth of Fusarium spp. [89]. However, it has been demonstrated that quercetin-glycoside was not as antifungal as its aglycone [90]. In contrast to quercetin, some other flavonoids have demonstrated strong antifungal effects as glycosides. For example, naringenin pyranoside demonstrated some antifungal activity with MIC values of 1600–3200 µg/mL against Candida albicans and C. krusei [91]. Therefore, naringenin-4′-methoxy-7-pyranoside, which we have found in the stem wood of T. brownii, is suggested to give some antifungal activity. Moreover, it has been found that flavonoids possessing methoxy groups are especially antifungal [87]. This would apply to naringenin-4′-methoxy-7-pyranoside (7) and 5,6-dihydroxy-3,4,7-trimethoxy flavone (12) which we have found in the stem wood of T. brownii. These flavonoids possess one and three methoxy groups, respectively, and thus are suggested to participate in the antifungal effects of the ethyl acetate extracts of the stem wood of T. brownii.

4. Materials and Methods

4.1. Collection of Plant Material

The stem wood and stem bark was collected from many individuals of Terminalia brownii growing in natural savannah woodland, in the Blue Nile Forest, in south-eastern Sudan (Figure 1). Voucher specimen were identified by the first author, Mr. Abdelazim Yassin Abdelgadir (Ph.D), Mr. Ashraf Mohamed Ahmed Abd Alla (Ph.D., Wood Sciences) and Mr. Haytham Hashim Gibreel (Ph.D., Taxonomy) at the Faculty of Forestry, University of Khartoum, Sudan and Mr. El Sheikh Abd alla Al Sheikh (Ph.D., Taxonomy) at Soba Forest Research Center, Khartoum, Sudan (Ph.D., Taxonomy). The Voucher specimens are deposited in the herbarium at the Department of Forest Products and Industries, Faculty of Forestry, University of Khartoum, Sudan.

4.2. Extraction

Hundred (100) grams of the dried and powdered stem wood and bark were used for the extractions. Extraction was initiated with sequential extraction, beginning with petroleum ether, followed by chloroform and finally the marc was extracted using 80% methanol. The 80% methanolic extract was subjected to liquid/liquid fractionation using ethyl acetate and this fractionation resulted in aqueous and ethyl acetate fractions.
4.3. Thin Layer Chromatography (TLC)

Using micro-capillary pipettes, 5 µL of ethyl acetate extracts (5 mg/mL) of the stem bark and wood of *T. brownii* were applied on normal phase silica gel thin layer plates (Kieselgel 60 F254, aluminum backed, Merck, Darmstadt, Germany) and on reversed phase thin layer plates (RP-18 F254s, Merck, Darmstadt, Germany) to detect compounds of a wide range of polarities. Toluene: ethyl acetate: formic acid (4:5:1, \( v:v:v \)) was used as an eluent for NP-TLC, while methanol: water: acetic acid (6:2:2) was used for RP-TLC. The development distance was 8 cm. The plates were sprayed with Vanillin-\( \text{H}_2\text{SO}_4 \), Dragendorff reagent, aluminum chloride and Natural Products reagents to detect various compound classes such as essential oils, terpenes, phenolic compounds, alkaloids and flavonoids [57]. The plates were observed in UV-light at 254 and 366 nm. A Camaq Video documentation system was used for photographing the plates.

4.4. Solid Phase Extraction (SPE)

LC-18 reversed phase cartridges (Supelco, Sigma-Aldrich, Darmstadt, Germany) were used for solid phase extraction in order to purify and enrich flavonoids and for separation of sugars and other interfering matrix compounds. The columns were equilibrated with 100% water and elution was performed using a gradient from 100% to 50% water followed by 100% methanol.

4.5. Reversed Phase High Performance Liquid Chromatography Coupled to Diode Array Detection (HPLC-UV/DAD)

The Agilent 1100 series HPLC system was used for the HPLC runs. The system consisted of an Agilent 1100 autosampler connected to Agilent series 1200 binary pump system coupled to an Agilent series 1100 thermostatic column compartment and an Agilent series 1100 DAD detector. Separations were performed on a reversed phase column (Varian LC-18; 4.6 mm × 250 mm; ID 5 µm, USA) at 30 °C and the flow rate was 0.5 mL/min. 5 µL of samples (5 mg/mL in 80% aqueous methanol) were injected. Gradient elution was performed using solvent (A) water +1% of acetic acid to increase peak resolution. Solvent (B) 100% acetonitrile. The step gradient began with 90% A and stopped while reaching 10% B in 30 min. After this 100% B was used for 5 min followed by 10% B for 5 min. Wavelengths of 254, 320, 360 and 380 nm were used for detection. The data was compared to standard compounds and computer libraries of pure compounds.

4.6. LC-Triple Quadrupole Mass Spectrometric Analysis (LC-MS and LC-MS/MS Tandem Mass Spectrometry)

An HPLC apparatus (1100 series, Agilent, Waldbronn, Germany) connected to an electrospray ionization (ESI) triple quadruple mass spectrometer (HTC Ultra-Bruker Daltonics-Advanced Mass Spectrometry Instrumentation, Germany) was used. Gradient elution was performed using acetonitrile (MeCN) and water containing 0.005% formic acid (Solvent A) and acetonitrile and glacial acetic acid (Solvent B). A linear gradient from 4% to 33% B was employed for 35 min and was increased to 100% B for 5 min. Then 4% B was used for 5 min to re-equilibrate. Mass analysis of compounds was performed using negative ion mode. The spray voltage was set to 5000 V and the capillary temperature to +280 °C. Nitrogen was used as sheathing gas and the flow was set to 40 U. Collision-induced dissociation (CID-MSn) was applied to induce fragmentation of the molecular ions, and their fragments were analyzed using tandem mass spectrometry. Helium was used as collision gas at 0.8 m Torr. Collision energies of 15 and 30 eV were used to investigate neutral loss and product ions and scanning was performed using a mass range from 50 to 1000 m/\( z \). Data from the literature, the Wiley Natural product library, and authentic samples were used for the structural identifications of phenolic compounds such as flavonoids, stilbenes and ellagic acid derivatives as well as triterpenes.
4.7. Antifungal Assays

4.7.1. Fungal Strains

*Aspergillus niger* ATCC 9763, *Nattrassia mangiferae* ATCC 96293, *Aspergillus flavus* ATCC 9763 and *Fusarium moniliforme* ATCC 24378 were obtained from National Research Center, Sudan. Before use, the strains were sub-cultured on Sabouraud dextrose agar (Oxoid™ CM0041B) slants, at +35 °C.

4.7.2. Agar Well Diffusion Method

A cup well agar diffusion method [55,56] with minor modifications was used. Before the test, the fungal strains were grown on petri dishes (Ø = 9 cm) containing Sabouraud dextrose agar at +35 °C overnight [92]. The resulting fungal growth was washed with 100 mL sterile normal saline to obtain fungal suspension containing conidia, which were used for the tests. 200 µL of this fungal suspension was adjusted to 1.0 × 10⁶ CFU/mL and mixed with 20 mL of sterile, molten Sabouraud dextrose agar which was poured into sterile petri dishes (Ø = 9 cm). The petri dishes were left to set at room temperature. Four holes were cut in the agar using a sterile cork borer (10 mm in diameter) and each hole was filled with 100 µL of extracts (1 mg/mL in 50% methanol) and amphotericin B (Sigma-Aldrich, 1 mg/mL in 50% methanol). 100 µL of 50% methanol was used as a negative control. The extracts/antibiotics/solvents were left to diffuse into the agar in the cold room (+4 °C) for one hour. The plates were then incubated at +35 °C for 24 h. For each experiment four replicates (n = 4) were used. The diameters of the zones of inhibition (IZ) were measured in mm using a caliper and the mean of five diameters ± SD and SEM was calculated.

4.7.3. Agar Dilution Method

Minimum inhibitory concentrations were determined using a slightly modified agar macrodilution method [93]. Fungal conidial suspensions were grown for four days in Sabouraud dextrose broth at +35 °C. For the test, 1 mL of these suspensions were diluted with 0.9% (w/v) NaCl to contain 1.0 × 10⁶ CFU/mL. 100 µL of these fungal suspensions were mixed with 10 mL of molten Sabouraud dextrose agar which was pipetted into a petri dish (Ø = 9 mm). 10 mL of twofold dilutions of plant extracts (from 500 to 31.25 µg/mL) and amphotericin B (from 500 to 15.625 µg/mL) were added to the petri dishes. Each dilution contained 500 µL of 50% methanol or hexane solutions of the plant extracts or antibiotics dissolved in 10 mL of molten Sabouraud dextrose agar. The petri dishes were incubated for 24 h at +35 °C. The MIC was taken as those concentrations that resulted in clear petri dishes showing no visible fungal growth. All tests were performed in triplicates. The solvents used for the plant extractions, 50% methanol or hexane, were used as negative controls. Hexane was used for dissolving those extracts which did not dissolve in 50% MeOH, that is very nonpolar extracts, such as those originating from hexane and petroleum ether extractions.

4.7.4. Statistical Analysis

The Student’s t-test provided by Microsoft Excel was used for the evaluation of the statistical significance of any differences between the antifungal results of the inhibition zones (IZ) of the tested extracts.

5. Conclusions

Ethyl acetate and aqueous extracts of the stem wood and bark of *Terminalia brownii* give good antifungal effects against *Nattrassia mangiferae*, *Fusarium verticillioides*, *Aspergillus flavus* and *Aspergillus niger*. Altogether twelve compounds were identified from an ethyl acetate extract of the stem wood of *T. brownii*. Cis- and trans-isomers of resveratrol 3-O-β-galloyl-glucoside were characterized for the first time in this species of *Terminalia*. Likewise, gallagic acid dilactone has not been reported previously in the stem wood of *T. brownii*. Owing to its relative chemical stability and its reported
antifungal efficiency against phytopathogenic molds, gallagic acid dilactone might be an especially interesting component in standardized antifungal extracts of T. brownii. Also, standardized extracts of T. brownii stem wood, enriched with ellagitannins, could be used as natural fungicides for protecting crops and as medicines to treat fungal infections. Ellagitannins purified from these extracts, if found to be more active than the extracts, could be used for ecological crop plant protection and wood preservation, while being relatively stable and possessing less toxicity than synthetic fungicides.

Our results provide partly the justification for the uses of water-based extracts of T. brownii for the protection of crop plants and for wood preservation in Africa, although phytochemical analysis of these aqueous extracts would be needed. Further studies are needed on the antifungal activities of separated compounds from both the aqueous and ethyl acetate extracts as well as on various controlled combinations of these compounds. In summary, standardized extracts of T. brownii stem wood could be used as new, cheaper and eco-friendly fungicides for routine use in Africa instead of toxic synthetic fungicides.

**Supplementary Materials:** The following are available online at www.mdpi.com/2079-6382/6/4/37/s1, (Slides 1–12): Tandem mass spectra of molecular ions and their fragment product ions resulting from selected compounds in a T. brownii ethyl acetate extract of the wood.

**Acknowledgments:** The first author is grateful for the financial support for this research article by University of Khartoum, Sudan. We would also like to thank Nickoli Kuhnert and his group, School of Engineering and Science, International University, Bremen, Germany for providing LC-MS/MS research facilities.

**Author Contributions:** Enass Y. A. Salih, performed the data collections, extractions, antifungal analysis, and preliminary chemical identifications and wrote the draft manuscript. Hiba A. Ali, managed the study and provided support in analytical chemistry. Pia Fyhrquist has given a major contribution for the drafting of this manuscript. Other authors, Abdelazim Y. Abdelgadir, Mustafa K. M Fahmi, Mai H. Alamin and Ashraf M. Ahmed were also involved in drafting the final manuscript. All authors have thoroughly revised the paper, read and approved the final manuscript.

**Conflicts of Interest:** The authors declare no conflict of interest. The founding sponsors had no role in the design of the study; in the collection, analyses, or interpretation of data; in the writing of the manuscript, and in the decision to publish the results.

**References**


62. Xia, B.; Bai, L.; Li, X.; Xiong, J.; Xu, P.; Xue, M. Structural analysis of metabolites of asiatic acid and its analogue madecassic acid in zebrafish using LC/IT-MSn. [CrossRef] [PubMed]


65. Oeirichs, P.B.; Pearce, C.M.; Zhu, J.; Filippich, L.J. Isolation and structure determination of terminalin A toxic analogue madecassic acid in zebrafish using LC/IT-MSn. [CrossRef] [PubMed]


84. Mierziak, J.; Kostyn, K.; Kulma, A. Flavonoids as important molecules of plant interactions with the environment. *Molecules* 2014, 19, 16240–16265. [CrossRef] [PubMed]


© 2017 by the authors. Licensee MDPI, Basel, Switzerland. This article is an open access article distributed under the terms and conditions of the Creative Commons Attribution (CC BY) license (http://creativecommons.org/licenses/by/4.0/).