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Cytotoxicity and effect on protease activity of copolymer extracts containing catechin

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ABSTRACT

Objective: To evaluate cytotoxicity and effect on protease activity of epigallocatechin-gallate extracted from experimental restorative dental copolymers in comparison to the control compound chlorhexidine. **Methods:** Copolymer disks were prepared from bis-GMA/TEGDMA (70/30 mol%) containing no compound (control) or 1% w/w of either epigallocatechin-gallate or chlorhexidine. MDPC-23 odontoblast-like cells were seeded with the copolymer extracts leached out into deionized water. Cell metabolic activity was quantified by 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT) assay at 24, 48, 72 h. Inhibition of protease activity by resin extracts was measured by a collagenolytic/gelatinolytic enzyme activity assay and gelatin zymography. Data for MTT and protease inhibition were analyzed using two-way ANOVA followed by Tukey or Bonferroni *post hoc* tests ($\alpha = 0.05$).

Results: The MTT revealed that at 72 h, extracts from control (16.7%) and chlorhexidine (22.3%) copolymers induced significant reduction in cell metabolism ($p < 0.05$). All copolymer extracts caused enzymatic inhibition in a dose dependent manner ($p < 0.01$). Even when highly diluted, epigallocatechin-gallate extract had a significant antiproteolytic activity ($p < 0.05$). Zymograms showed that all extracts reduced activity of MMP-2 and MMP-9 (pro- and active forms), with MMP-9 exhibiting the highest percentage inhibition revealed by densitometry.

Conclusions: Epigallocatechin-gallate and chlorhexidine extracts did not exert cytotoxicity on evaluated cells when compared to control extracts. Both compounds retained antiproteolytic activity after extraction from a dental copolymer.

Clinical significance: Once extracted from a dental copolymer, epigallocatechin-gallate is not cytotoxic and retains antiproteolytic activity. These results may allow incorporation of epigallocatechin-gallate as a natural-safe alternative to chlorhexidine in functionalized restorative materials.

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1. Introduction

Resin composites are used in a variety of applications in dentistry including restorations, liners, cores, cementation, and endodontic sealers. The composition of resins has evolved

significantly since they were introduced, first aiming to improve wear resistance and polish, lower the polymerization shrinkage and later to develop self-adhesive materials (Chen, 2010). However, secondary caries (Rasines Alcaraz et al., 2014) and loss of bond strength due to proteolytic activity of collagenolytic and gelatinolytic enzymes in the hybrid layer (Tjäderhane et al., 2013), still remain to be the problems reducing the composite filling longevity. Enzymes such as MMP-2 and MMP-9, localized in dentin or saliva, may have a fundamental role in dentin organic matrix degradation and caries progression (Tjäderhane et al., 1998; Mazzoni et al., 2007). Both MMP-2 and MMP-9 were indeed found to be among the most abundant proteases found in caries-affected dentin (Vidal et al., 2014). Future developments should

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thus aim to produce functionalized materials, with therapeutic activities, such as antibacterial and/or antiproteolytic properties (Tjäderhane et al., 2013; Jandt & Sigusch, 2009; Delaviz, Finer, & Santerre, 2014).

Chlorhexidine is capable of arresting caries when applied to dentin (Van Rijkom, Truin, & vant Hof, 1996). Chlorhexidine also inhibits dentin endogenous proteases such as MMPs and cysteine cathepsins (Tjäderhane et al., 2013). This antiproteolytic effect prevents hybrid layer degradation and increases dentin bond strength durability (Tjäderhane et al., 2013), and may also participate in the prevention of dentinal caries progression (Garcia et al., 2009). The release of chlorhexidine from resins by Fickian diffusion has been widely tested in terms of mechanical and biological properties (Leung et al., 2005; Anusavice, Zhang, & Shen, 2006; Pallan, Furtado Araujo, Cilli, & Prakkki, 2012), confirming the antibacterial efficacy also when incorporated into adhesive resin (Hiraishi, Yiu, King, Tay, & Pashley, 2008). However, certain drawbacks of this compound such as its synthetic nature and a possible cytotoxicity to dentin odontoblastic cells (Lessa, Nogueira, Huck, Hebling, & Costa, 2010; de Souza, de Aquino, de Souza, Hebling, & Costa, 2007) motivate researchers to investigate alternatives.

Epigallocatechin-gallate (EGCg), a natural polyphenol from green tea, has been shown to be effective antimicrobial against *Streptococcus mutans* and in inhibiting acid production in dental biofilm (Hirasawa, Takada, & Otake, 2006; Xu, Zhou, & Wu, 2012). Incorporation of EGCg has also been shown not to adversely affect the mechanical properties of different copolymer compositions (Pallan et al., 2012), and EGCg released from these copolymers retains antibacterial activity against *S. mutans* (Mankovskaia, Levesque, & Prakkki, 2013). Moreover, it has been reported that EGCg inhibits dentinal proteases (Kato et al., 2012), thus preserving the long-term dentin bond strength with an effect equal to chlorhexidine (Santiago, Osorio, Neri, Carvalho, & Toledano, 2013). However, the protease activity of compounds released from dental copolymers has not yet been demonstrated. It was therefore the objective of this study to investigate the cytotoxicity and protease-inhibiting activities of EGCg, released by an experimental dental copolymer, and compare that to chlorhexidine. The working null hypotheses were (1) extracts from EGCg and chlorhexidine incorporated copolymers will not exert cytotoxicity against MDPC-23 odontoblast-like cells, and (2) extracts from EGCg and chlorhexidine incorporated copolymers will not exhibit collagenolytic and gelatinolytic activities.

2. Materials and methods

Chemicals were obtained from Sigma–Aldrich (St. Louis, MO, USA) unless otherwise specified.

2.1. Formulation of copolymers and compound extraction

Experimental copolymer disks (7 mm diameter × 2 mm high) were prepared from bis-GMA (bisphenol glycidyl dimethacrylate) and TEGDMA (triethyleneglycol dimethacrylate) at 70/30 mol% ratio, containing no compound (control) or with 1% w/w of either epigallocatechin-gallate or chlorhexidine diacetate. The 1% compound ratio was chosen based on previous study that reported stability in mechanical properties of same copolymer incorporated with 1% of either epigallocatechin-gallate or chlorhexidine diacetate (Pallan et al., 2012). Resins were activated for visible light polymerization (Demi LED, Kerr Co., Middleton, USA; 540 mW/cm²) by the addition of camphorquinone and 2-(dimethylamino) ethyl methacrylate (0.2% w/w each) (Pallan et al., 2012), and stored at 6 °C until use.

A UV–vis spectrophotometer (Synergy HT Multi-Mode Microplate Reader, BioTEK, Winooski, VT, USA) was used to analyze the compound release rates ($n = 3$) after 24 h storage in 1 mL deionized water (EGCg OD₂₉₇ and chlorhexidine OD₂₅₇) at 37 °C. For the following tests, the copolymer extracts were concentrated 10× in volume (Eppendorf Concentrator Plus, Eppendorf AG, Hamburg, Germany).

2.2. MDPC-23 odontoblast-like cell culture

The immortalized mouse dental papilla MDPC-23 cells were maintained in Dulbecco's Modified Eagle's medium (DMEM; with high glucose, L-glutamine and sodium pyruvate) supplemented with 10% fetal bovine serum (Gibco, Grand Island, NY, USA), with 100 IU/mL penicillin, 100 µg/mL streptomycin and 0.25 µg/mL fungizone (Gibco). The cells were sub-cultured every 3 days and allowed to grow in a humidified incubator at 37 °C with 5% CO₂ (Isotemp; Fisher Scientific, Pittsburgh, PA, USA). Studies have shown that MDPC-23 synthesizes dentin-specific proteins (i.e., dentin sialoprotein) that are synthesized mainly by odontoblasts. These findings support the idea that MDPC-23 may serve as a valuable *in vitro* model for studies of functional roles of odontoblasts (Shin, Yeon, Oh, & Kim, 2006).

2.3. MTT cytotoxicity assay

MDPC-23 cells were seeded in DMEM in 96-well plate at 3×10^3 cell/200 µL/well and allowed to grow for 72 h. At day 3, the new medium was incorporated with the copolymer extracts diluted at 40 µL/100 µL. The cells were continuously cultured and evaluated at 24, 48 and 72 h cell growth time points ($n = 3$). The extract from control copolymer was used as positive control and PBS (Phosphate Buffered Saline) as negative control. Cells metabolic activity was evaluated by the MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) assay (Roche Applied Science, Indianapolis, IN, USA) as previously described (Aranha, Giro, Hebling, Lessa, & Costa, 2010). Data were analyzed by two-way ANOVA and Tukey's *post hoc* test ($\alpha = 0.05$).

2.4. Gelatinolytic/collagenolytic activity assay

The effect of copolymer extracts on functional enzyme activity was assayed using EnzChek gelatinolytic/collagenolytic assay kit (Molecular Probes; Eugene, OR, USA) supplemented with type IV *Clostridium histolyticum* collagenase (Molecular Probes). The fluorescence of the collagen substrate is internally quenched and is only released when it is cleaved enzymatically into highly fluorescent, low molecular weight peptides. Different dilutions of copolymer extracts (80 µL to 1 µL) were mixed with quenched fluorescent substrates in a final volume of 200 µL of reaction buffer in 96-well plates ($n = 6$). The rate of proteolysis was determined by a Synergy™ Mx Monochromator-Based Multi-Mode Fluorometer (BioTek), operated at an absorption maxima of 495 nm and fluorescent emission maxima of 515 nm. The assay included a series of control collagenase standards as well as reagent blanks. The change in enzymatic activity by EGCg or chlorhexidine extracts was calculated (%) in comparison to the control extracts within each tested dilution. Data were analyzed by two-way ANOVA and Bonferroni's *post hoc* test ($\alpha = 0.05$).

2.5. Zymographic analysis

MMP-2 and MMP-9 were in-house gelatin-Sepharose column-purified from conditioned cell culture media as previously described (Mäkelä, Salo, Uitto, & Larjava, 1994). Aliquots of purified human MMP-2 (9 µL + 3 µL deionized water) or MMP-9 (12 µL)

Table 1

Cytotoxicity (OD₅₇₀) of experimental resin extracts towards MDPC-23 odontoblast-like cells. The lower values indicate reduction in cell viability.

Groups	Time point		
	24 h (s.d.)	48 h (s.d.)	72 h (s.d.)
No resin extract (PBS)	1.7 (0.20) ^{Aa}	2.2 (0.10) ^{Ba}	1.8 (0.10) ^{Aa}
Control resin extract	1.8 (0.07) ^{Aa}	2.1 (0.13) ^{Ba}	1.5 (0.11) ^{Cb}
EGCg resin extract	1.8 (0.08) ^{Aa}	2.1 (0.17) ^{Ba}	1.5 (0.24) ^{Aa,b}
Chlorhexidine resin extract	1.6 (0.14) ^{Aa}	2.0 (0.11) ^{Ba}	1.4 (0.17) ^{Ab}

Two-way ANOVA and Tukey, $n = 3$, OD₅₇₀ = optical density at 570 nm, s.d.: standard deviation, $\alpha = 0.05$. PBS: phosphate buffered saline. Same upper-case letters indicate no statistical difference within each row. Same lower-case letters indicate no statistical difference within each column.

were incubated with 3 μ L of the different copolymer extracts for 2 h at 37 °C, and diluted into 4 μ L of Laemmli sample buffer. These were subjected to electrophoresis under non-reducing conditions in 10% sodium sulphate-polyacrylamide gel (SDS-PAGE) containing 1% type I bovine soluble skin gelatin. Pre-stained SDS-PAGE standards (BIO-RAD Laboratories Ltd., Hercules, CA, USA) were used as low-range molecular-weight markers. After electrophoresis (120 V/25A), the gels were incubated for 30 min in buffer I (50 mmol/L Tris–HCl, 2.5% Tween 80, 0.02% Na₂S₂O₄ (w/v) at 22 °C pH 7.5). Subsequently, the gels were incubated again for 30 min in buffer II (50 mmol/L Tris–HCl, 2.5% Tween 80, 0.02% Na₂S₂O₄ (w/v), 1 μ M ZnCl₂, and 5 mmol/L CaCl₂ at 22 °C). Afterwards, the gels were incubated for 12 h in buffer III (50 mmol/L Tris HCl, 5 mmol/L CaCl₂, 1 μ M ZnCl₂, and 0.02% Na₂S₂O₄ (w/v) at 37 °C). Gels were stained in 0.1% Coomassie Brilliant Blue R-250 for 60 min at 22 °C and destained in a solution containing 10% acetic acid and 10% methanol for 30 min. The gels were then scanned (Imagescanner, Amersham Biosciences, Uppsala, Sweden) and band intensities quantified with Kodak molecular imaging software (version 4.5; Kodak, Rochester, NJ, USA). The inhibition of MMP activities by EGCg or chlorhexidine copolymer extracts was calculated (%) in comparison to the control extracts within each gel ($n = 3$).

3. Results

The mean compound release rates were $0.94 \pm 0.008 \mu\text{g}/\text{cm}^2$ for EGCg and $2.18 \pm 0.191 \mu\text{g}/\text{cm}^2$ for chlorhexidine. Cytotoxicity data are shown in Table 1. The MTT assay revealed that at 24 h ($p = 0.07$) or 48 h ($p = 0.37$) cell growth, test groups did not differ statistically from each other or from the negative control (PBS). At 72 h, control and chlorhexidine copolymer extracts significantly reduced cell metabolism ($p < 0.05$) as compared to PBS (16.7% and 22.3%, respectively). EGCg extract did not significantly differ from any other group.

Data for gelatinase/collagenase activity are shown in Table 2. Both EGCg and chlorhexidine extracts inhibited type IV C. *histolyticum* collagenase in a dose-dependent manner ($p < 0.0001$). When compared to the control, the chlorhexidine extract significantly inhibited enzymatic activity in all tested dilutions,

except for the lowest dilution of 1 μ L/200 μ L reaction buffer. All EGCg extract dilutions significantly inhibited enzymatic activity when compared to the control ($p < 0.05$).

The zymographic assays revealed four major bands. Two bands with approximate molecular mass of 72 kDa and 66 kDa correspond to pro-MMP-2 and MMP-2 active forms, respectively. Two other bands with approximate molecular mass of 95 kDa and 80 kDa correspond to pro-MMP-9 and MMP-9 active forms, respectively. These bands were confirmed to be MMPs as their activities were completely inhibited by 1,10-phenanthroline, a specific MMP inhibitor (data not shown). All copolymer extracts in the increasing percentage order of control, EGCg and chlorhexidine, reduced gelatinolytic activities of pro- and active MMP-2 and MMP-9, with the latter exhibiting higher percentage inhibition as revealed by densitometry (Figs. 1a and 1b).

4. Discussion

When testing different copolymer extracts towards MDPC-23 cells, our results were in line with Yasuda et al. (Yasuda et al., 2008) who reported cytotoxicity caused by polymerized resinous agents to be time dependent and detectable at 72 h cell growth. At 72 h, extracts from control copolymer induced significant reduction of 16.7% in cell metabolism. Leachable components of copolymers include unreacted monomers, oligomers and free radical species that are associated with several forms of tissue damage (Kunawarote et al., 2010). Also at 72 h, chlorhexidine copolymer extract led to cell metabolism reduction of 22.3%. The dissociation of chlorhexidine salts in positively charged ions may have caused additional release of low molecular weight components from the cell wall, as these ions adsorb to the cell wall causing alterations in the osmotic balance (Lessa et al., 2010).

The fact that the cytotoxicity of EGCg extract did not statistically differ from any other group suggests that the observed reduced cell metabolism (16.7%, similar to positive controls) was mainly caused by the copolymer by-products. Nonetheless, the lower EGCg release rates than for chlorhexidine should be taken into consideration. Although EGCg (~5 mg/ml, pH 7) is more soluble in water than chlorhexidine (~3.5 mg/ml, pH 7), diffusion of EGCg through copolymer chains is probably prevented by its more branched polyphenolic molecular structure and by a possible hydrogen bonding through bis-GMA pendant hydroxyl groups (Pallan et al., 2012). Catechins have also been recognized to have excellent chemopreventive properties, including protection of normal cells against genotoxic effects (Roy, Chakrabarty, Sinha, Bhattachary, & Siddiqi, 2003). Antioxidants like EGCg are capable of neutralizing free radical species, as reactive oxygen can abstract hydrogen atoms from EGCg (Kunawarote et al., 2010). However, the *in vitro* cytotoxicity findings do not directly infer *in vivo* conditions. Many factors contribute in decreasing the cytotoxic effects of dental material leachable components in vital teeth, such as diffusion barrier and intermolecular interactions with dentin tissue, positive intrapulpal pressure, and the presence of outward

Table 2

Inhibition of proteolytic activity as observed with the EnzChek gelatinase/collagenase assay. The values, expressed in U/mL, demonstrated the enzyme activity.

Solution subfractions (μ L/200 μ L)	Control resin extract (s.d.)	EGCg resin extract (s.d.)	Chlorhexidine resin extract (s.d.)
1	0.197 (0.008) ^{Aa}	0.151 (0.012) ^{Ba}	0.178 (0.012) ^{Aa}
5	0.192 (0.005) ^{Aa}	0.143 (0.008) ^{Ba}	0.153 (0.015) ^{Bb}
10	0.182 (0.006) ^{Aa}	0.113 (0.041) ^{Ba}	0.130 (0.013) ^{Bb}
40	0.129 (0.037) ^{Ab}	0.103 (0.011) ^{Bb}	0 ^{Cc}
80	0.103 (0.011) ^{Ac}	0.036 (0.040) ^{Bc}	0 ^{Cc}

Two-way ANOVA and Bonferroni, $n = 6$, s.d.: standard deviation, $\alpha = 0.05$. Same upper-case letters indicate no statistical difference within same row. Same lower-case letters indicate no statistical difference within same column.

Fig. 1a. Zymographic analysis of purified human MMPs treated with bis-GMA/TEGDMA copolymer extracts. *Lanes 1 and 2:* purified human MMP-2 (pro- and active forms at 72 and 66 kDa) and MMP-9 (pro- and active forms at 92 and 86 kDa), respectively. *Lanes 3 and 4:* MMP-2 and MMP-9 incubated for 2 h (3:1 and 4:1 v/v) with control copolymer extracts (no compounds added), respectively. *Lanes 5 and 6:* MMP-2 and MMP-9 incubated for 2 h (3:1 and 4:1 v/v) with EGCg copolymer extracts, respectively. *Lanes 7 and 8:* MMP-2 and MMP-9 incubated for 2 h (3:1 and 4:1 v/v) with chlorhexidine (CHX) copolymer extracts, respectively. It can be noted that MMP-9 (pro- and active forms) had higher inhibition as compared to MMP-2 (pro- and active forms).

fluid flow and cytoplasmic processes of odontoblasts in dentin tubules (Lessa et al., 2010).

Of prospective benefit to the prevention of caries progression, our results confirmed the inhibitory interaction of EGCg and chlorhexidine extracts on gelatinolytic/collagenolytic proteases that can be found in dentin, more specifically to MMP-2 and MMP-9. The present results also showed over two-fold inhibition of MMP-9 compared to MMP-2. Similar results were previously reported by Demeule, Brossard, Page, Gingras and Beliveau (2000). In that study, the low IC_{50} values obtained for EGCg (1 μ M) with MMP-9 actually suggested that these MMPs could be targets for green tea polyphenols since the EGCg plasma concentration is reported to be $\sim 0.2 \mu$ M in humans after few cups of tea (Cao & Cao, 1999; Yang, Liao, Kim, & Yurkow, 1998). Cao & Cao Nature 1999 (Demeule 34), Yang et al. Carcinogenesis 1998 (Demeule 35). Not much is known about the molecular mechanisms by which EGCg blocks MMP activities. It has been suggested that inhibition of gelatinases by EGCg may be due to zinc chelation, as polyphenols have high affinity to metal ions (Yang et al., 1998), and zinc is essential for enzymatic activity. Alternatively, Garbisa et al. (Garbisa et al., 2001) suggest a direct EGCg-protein interaction model that is described according to the following three equilibria: I) MMP interacts with EGCg with high affinity constant; II) gelatin interacts with EGCg; III) MMP-EGCg interacts with gelatin leading

to the formation of a ternary inactive complex (MMP-EGCg + gelatin $\leftrightarrow$ MMP-EGCg-gelatin). At high EGCg/MMP ratios, gelatin-bound enzyme recovery decreases, indicating that the free flavanol competes with the flavanol-MMP complex for binding to gelatin. Since among all green tea catechins ECg (epigallocatechin), followed by EGCg, are shown to be up to 30 $\times$ more effective MMP inhibitors than other catechins indicates that the presence of flavanol skeleton with galloyl radical is necessary for producing the inhibitory activity (Demeule et al., 2000).

It has been reported that a chelating mechanism is also involved in the inhibitory mechanism of MMP-2 and MMP-9 by chlorhexidine. This is because the addition of calcium chloride to the assay mixture was shown to prevent the inhibition of these enzymes (Gendron, Grenier, Sorsa, & Mayrand, 1999). In this present study, compound release rates in 1 mL deionized water were above the minimum required for EGCg (MMP-2 and MMP-9: 0.00075%) and chlorhexidine (MMP-2: 0.0001% and MMP-9: 0.002%) to inhibit MMPs (Gendron et al., 1999; Yamamoto et al., 2003). However, when comparing our results with previously published results (Gendron et al., 1999; Yamamoto et al., 2003), the main difference noted was that in our study, although with higher compound rates, lower inhibition values were observed. We can only speculate about a possible explanation for our findings. The above-mentioned leachable components of bis-GMA/TEGDMA

Fig. 1b. Densitometric evaluation of zymographic bands expressed as percentage change of MMP-2 and MMP-9 activities after treatment with different resin extracts.

copolymer extract are possibly interacting with EGCg flavanol and dissociated chlorhexidine ions, as well competing for binding with enzymes. The assumption is based on our zymographic results (Figs. 1a and 1b), showing that the control copolymer extract also caused slight but undeniable inhibition to MMP-2 and MMP-9 (pro- and active- forms). Moreover, the interplay between components eluted by the copolymer and consequent availability of active MMP inhibitory species, along with the affinity of EGCg and chlorhexidine for collagenolytic enzymes, might explain EGCg extract to inhibit collagenolytic enzyme activity even at lower concentration than chlorhexidine extract, as shown by the collagenolytic/gelatinolytic enzymatic assay. Indeed, the inhibition of MMP-2 by low concentrations (0.62%) of pure TEGDMA monomer has been demonstrated (de Carvalho et al., 2011). The molecular mechanisms are believed to rely on complexes formed between metal ions and the monomer crown ethers. Instead, it could also be due to a Lewis acid-base complexation with the monomer carbonyls, or finally, a reaction between the monomer carbonyl moieties with the nucleophilic centers on MMPs outside catalytic domain (de Carvalho et al., 2011).

In experiments testing the inhibition of proteolysis using EnzChek, collagenase from *C. histolyticum* is used. Even though the collagen degradation sites of bacterial collagenase are different from mammalian enzymes, studies indicate that it can be used as a model in *in vitro* inhibition experiments related to dental tissues. Chlorhexidine (Kato et al., 2012; Hiraishi et al., 2011; Zheng, Hu, Chen, Zhu, & Chen, 2012) and EGCg (Kato et al., 2012) have both been shown to inhibit *C. histolyticum* collagenase function in demineralized dentin matrix. Together with the MMP inhibition observed in this study, the EnzChek data support the antiproteolytic activity of chlorhexidine and EGCg extracts. In fact, inhibition by EGCg was seen even for the lowest volume tested, but complete inhibition could not be observed even for the highest volumes. Contrarily, the lowest volume of chlorhexidine did not significantly inhibit the proteolytic activity when compared with control, but the degree of inhibition was much more pronounced for the highest volumes employed and complete inhibition was achieved. In conclusion, the first proposed null hypothesis was accepted as EGCg and chlorhexidine copolymer extracts did not cause significant cytotoxicity on dentin cells when compared to the control extract. Both compounds retained antiproteolytic activity after extraction from a dental copolymer, rejecting the second null hypothesis. In order to improve therapeutic copolymer formulations, further studies are required to determine specific molecular interactions between copolymer leachable components and the incorporated compounds.

Conflict of interest

The authors declare that there are no conflict of interest involving this article.

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Ethical approval

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