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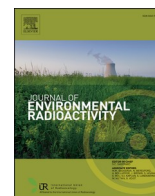
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## Transfer of transuranium elements along the food chain lichen-reindeer-man – A review of investigations in Finnish Lapland

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### ABSTRACT

Following the atmospheric nuclear tests in the '50s and early '60s radioecological research on the (sub)arctic food chain lichen-reindeer/caribou-man was initiated in Finland among other northern countries. The enrichment of radionuclides in this food chain can lead to exceptionally high body burdens among the indigenous Sami and Inuit populations consuming large quantities of the meat and edible organs of reindeer and caribou.

In Finland, first fission and activation products and natural radionuclides were studied but in the early 1970s the investigations concerning transuranium elements were started. These studies have continued to the present as also the effects of the Chernobyl accident on the existence of neptunium, plutonium, americium and curium isotopes in the environment of northern Finland have been investigated. In addition to radioactivity measurements detailed dietary surveys were performed among the reindeer herders and other Sami persons to assess the human intake of radionuclides by ingestion.

The main aim of this literature review is to summarize the obtained data concerning transuranium elements in the food chain lichen-reindeer-man in northern Finland but also some supporting data is included.

### 1. Introduction

Considerable amounts of man-made radioactivity have been transported to the Arctic environment since the start of the nuclear era in 1945. There has been a variety of sources contributing to this: nuclear weapons tests, the 1986 Chernobyl accident, liquid and atmospheric emissions from the nuclear fuel reprocessing plants in Sellafield and Dounreay, the United Kingdom, La Hague, France, operational and accidental liquid and atmospheric emissions from the nuclear weapons production and fuel reprocessing facilities in Russia (Chelyabinsk, Tomsk and Krasnoyarsk), Soviet dumping of nuclear waste and used reactors to the Arctic Ocean. Local contamination has occurred due to accidents involving aircrafts carrying nuclear weapons, nuclear-powered vessels and the handling of their fuel cycle. Especially important of these has been the Novaya Zemlya nuclear test site. The Soviet Union conducted there 91 nuclear tests in the atmosphere, ground surface and water in 1955–1962 including the most powerful atmospheric nuclear test ever (“Tsar Bomba” or “Big Ivan”, 50–58 megatons) on October 30, 1961 (Arctic Monitoring and Assessment Programme, 2010). The most recent significant source of artificial radioactivity transported to the Arctic is the 2011 Fukushima accident (Paatero et al., 2012).

Finnish Lapland has received radioactivity from the nuclear weapons tests, Chernobyl and Fukushima nuclear accidents and the 1964 SNAP-9A satellite re-entry accident (Krey, 1967). The radionuclides deposited included also transuranium elements plutonium, americium, curium, and neptunium. The environmental behaviour of these elements is important to know for several reasons. Most of them are alpha emitters and thus highly radiotoxic. Many of them have long physical and biological half-lives. And finally, large amounts of these elements are formed in nuclear explosions and during the operation of nuclear reactors.

Following the atmospheric nuclear tests in the '50s and early '60s several radioecological research projects, focused on the (sub)arctic food chain lichen-reindeer/caribou-man, were initiated in Scandinavia and North America (Miettinen et al., 1963; Svensson and Lidén, 1965; Hanson, 1967; Holm & Persson, 1975, 1978a). Lichen collects deposited radionuclides efficiently and is the main fodder for reindeer during the winter season (Tuominen and Jaakkola, 1973). In summer and autumn reindeer eat mostly grass, birch leaves and mushroom (Rissanen and Rahola, 1989). The enrichment of radionuclides in this food chain can lead to exceptionally high body burdens among the indigenous Sami and Inuit (Eskimo) populations consuming large quantities of the meat and

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**Table 1**  
Radioanalytical methods used for determining transuranium radionuclides in Finnish radioecological studies.

Transuranium element	Separation method	Detection method	Reference
Pu	ashing and/or acid digestion + anion exchange	Alpha spectrometry ( $^{238}\text{Pu}$ , $^{239}\text{Pu}$ , $^{240}\text{Pu}$ )	Tulikoura et al. (1974); Jaakkola et al. (1975); Miettinen (1976); Keinonen et al. (1977); Miettinen et al. (1980); Hakanen (1981); Jaakkola et al. (1981); Hakanen et al. (1984); Mussalo-Rauhamaa (1981); Mussalo-Rauhamaa et al. (1984)
		Liquid scintillation counting ( $^{241}\text{Pu}$ ) Alpha spectrometry ( $^{238}\text{Pu}$ , $^{239}\text{Pu}$ , $^{240}\text{Pu}$ )	Hakanen et al. (1984) Paatero et al. (1998); Paatero and Jaakkola (1998)
	acid digestion + calcium oxalate and iron hydroxide co-precipitations + anion exchange	Liquid scintillation counting ( $^{241}\text{Pu}$ ) SF-ICP-MS ( $^{240}\text{Pu}$ , $^{239}\text{Pu}$ )	Paatero et al. (1994) Salminen-Paatero et al. (2012)
		Alpha spectrometry ( $^{238}\text{Pu}$ , $^{239}\text{Pu}$ , $^{240}\text{Pu}$ )	Salminen and Paatero (2009); Salminen-Paatero et al. (2019)
Am + Cm	ashing and/or acid digestion + ion exchange + solvent extraction	Liquid scintillation counting ( $^{241}\text{Pu}$ ) SF-ICP-MS ( $^{240}\text{Pu}$ , $^{239}\text{Pu}$ )	Salminen-Paatero et al. (2012)
		Alpha spectrometry ( $^{241}\text{Am}$ )	Miettinen et al. (1980); Jaakkola et al. (1981)
	ashing and/or acid digestion + calcium oxalate and iron hydroxide co-precipitations + anion exchange in methanol solution	Alpha spectrometry ( $^{241}\text{Am}$ , $^{242}\text{Am}$ , $^{243}\text{Am}$ , $^{244}\text{Cm}$ )	Paatero et al. (1998); Paatero and Jaakkola (1998)
		Alpha spectrometry ( $^{241}\text{Am}$ , $^{242}\text{Am}$ , $^{243}\text{Am}$ , $^{244}\text{Cm}$ )	Salminen et al. (2005)
	ashing + acid digestion + calcium oxalate and iron hydroxide co-precipitations + anion exchange + extraction chromatography	Alpha spectrometry ( $^{241}\text{Am}$ )	Salminen et al. (2009)
ashing + acid digestion + extraction chromatography	Alpha spectrometry of old Pu alpha samples, for determining $^{241}\text{Pu}$ via ingrowth of $^{241}\text{Am}$ SF-ICP-MS ( $^{237}\text{Np}$ )	Harva (1976); Salminen-Paatero et al. (2014) Salminen et al. (2009)	
Np	ashing + acid digestion + oxalate precipitation + extraction chromatography	Gamma spectrometry ( $^{239}\text{Np}$ )	Rantavaara (1987)

edible organs of reindeer and caribou.

The Department of Radiochemistry, University of Helsinki, started the studies on lichen-reindeer-man food chain in the beginning of the 1960s. First fission products were studied (Salo et al., 1963; Miettinen and Häsänen, 1967) but when the United States Atomic Energy Commission (AEC) started to finance the studies also activation products (Jaakkola, 1969), natural radionuclides (Kauranen and Miettinen, 1969) and, in the early 1970s, transuranium nuclides were investigated (Tulikoura et al., 1974; Jaakkola et al., 1975, 1978). Altogether AEC and its successors, United States Energy Research and Development Administration and United States Department of Energy, funded these studies nicknamed “American project” for thirteen years, a rare case in the Finnish history of science. In addition to radioactivity measurements detailed dietary surveys were performed to assess the human intake of radionuclides by ingestion (Jokelainen et al., 1962; Jokelainen, 1965; Hasunen and Möttönen, 1976).

The main aim of this literature review is to summarize the obtained data concerning transuranium elements in the food chain lichen-reindeer-man in northern Finland but also some supporting data is included. A brief summary of the radioanalytical methods used in these studies is presented in Table 1.

## 2. Deposition of transuranium elements

According to Paatero et al. (2010) the strontium-90 ( $^{90}\text{Sr}$ ) global fallout from the atmospheric nuclear test was quite evenly distributed in Finland with only a slight downward trend from south to north. The deposition represented also well the zonal  $^{90}\text{Sr}$  deposition between the 60th and the 70th latitude. The deposition of  $^{90}\text{Sr}$  in northernmost Lapland was about 1 kBq/m<sup>2</sup>. Based on the measured ratio of plutonium-239,240 ( $^{239,240}\text{Pu}$ ) to  $^{90}\text{Sr}$ , 0.01763, it can be estimated that the  $^{239,240}\text{Pu}$  deposition was about 20 Bq/m<sup>2</sup>. On the other hand, Hardy

et al. (1973) give a zonal  $^{239,240}\text{Pu}$  deposition value of about 60 Bq/m<sup>2</sup> between the 60th and the 70th latitude and 13 Bq/m<sup>2</sup> between the 70th and the 80th latitude, in other words, of the same order of magnitude (Fig. 1). The  $^{239,240}\text{Pu}$  deposition from the 1986 Chernobyl accident was about 0.1 Bq/m<sup>2</sup> in Inari (Paatero et al., 1998). Only Halla, the southernmost reindeer herding district in Finland, east of Kajaani next to the Finnish-Russian border (Fig. 2), received considerable amounts of deposited radionuclides from the Chernobyl accident.

Assuming that the  $^{239,240}\text{Pu}$  deposition due to nuclear tests was 20 Bq/m<sup>2</sup> we can calculate, based on activity ratios that the deposition of neptunium-237 ( $^{237}\text{Np}$ ),  $^{238}\text{Pu}$ , and  $^{241}\text{Pu}$  was 0.07, 0.6 and 320 Bq/m<sup>2</sup>, respectively, in northern Lapland. The above mentioned burning of the SNAP-9A  $^{238}\text{Pu}$  power source after the satellite re-entry increased the  $^{238}\text{Pu}$  deposition to 1.3 Bq/m<sup>2</sup>. The Chernobyl accident increased the  $^{237}\text{Np}$ ,  $^{241}\text{Am}$ ,  $^{238}\text{Pu}$  and  $^{241}\text{Pu}$  deposition with 0.0000125, 0.045, 0.05 and 9.5 Bq/m<sup>2</sup>, respectively (Paatero et al., 1994, 2002; Salminen et al., 2009). The Chernobyl-derived  $^{244}\text{Cm}$  deposition north of the Arctic Circle was less than 0.023 Bq/m<sup>2</sup> (Salminen et al., 2005).

## 3. Plutonium

The activity concentration of plutonium isotopes 239 and 240 in lichen in Lapland was about 4 Bq/kg dry weight (d.w.) in 1960–1962 (Fig. 3). Due to the intense nuclear testing just before the Partial Test Ban Treaty in 1963 the concentration level rose to 9 Bq/kg but returned to the 4 Bq/kg level in 1965. This level was maintained for several years by the atmospheric nuclear tests of the People’s Republic of China. Following the Chernobyl accident the  $^{239,240}\text{Pu}$  activity concentration in lichen was 0.306 Bq/kg d.w. in 1986 and 0.07 Bq/kg d.w. one year later (Paatero et al., 1998). The activity concentration of  $^{241}\text{Pu}$  varied between 110 and 1.5 Bq/kg d.w. in lichen between 1960 and 1976. In 1987 the  $^{241}\text{Pu}$  activity concentration had dropped below the detection limit

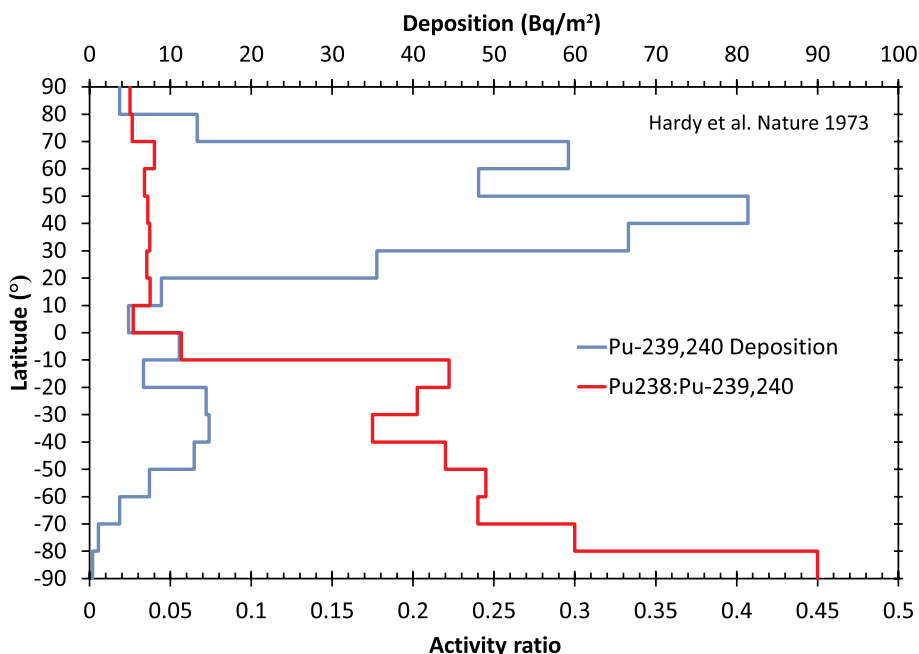


Fig. 1. Deposition of  $^{239,240}\text{Pu}$  and  $^{238}\text{Pu}/^{239,240}\text{Pu}$  activity ratio as a function of latitude (Data from Hardy et al., 1973). Finland is located between the 60th and the 70th latitude North.

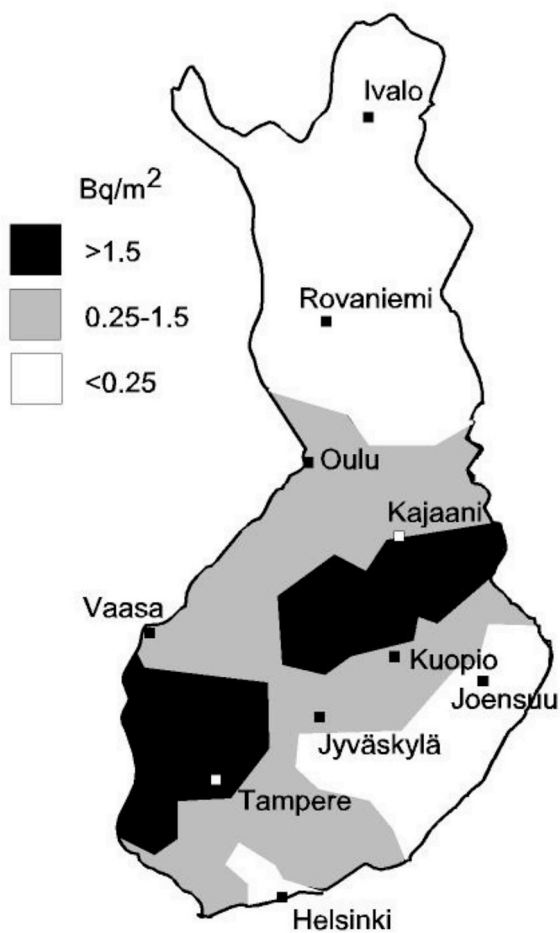


Fig. 2. Deposition of  $^{239,240}\text{Pu}$  in Finland after the 1986 Chernobyl accident (Paatero et al., 2006).

of 1.4 Bq/kg. For comparison, a lichen sample from south-western Finland contained  $5.7 \pm 0.3$  Bq/kg d.w. of  $^{239,240}\text{Pu}$  and another sample  $204 \pm 11$  Bq/kg d.w. of  $^{241}\text{Pu}$  (Hakanen et al., 1984; Paatero et al., 1994). A similar temporal behaviour was observed with birch leaves and grass which are important reindeer fodder in summer (Fig. 4). The maximum concentration level in tree lichen (*Alectoria* sp.), 17 Bq/kg d. w., was observed in 1966, in other words, three years later than the maximum deposition occurred. This may be related to the slow accumulation of radionuclides into tree lichen.

The  $^{238}\text{Pu}/^{239,240}\text{Pu}$  activity ratio was around 3% in the early 1960s, a typical value for plutonium originating from global weapons test fallout (Fig. 5). Owing to the 1964 SNAP-9A accident the activity ratio increased gradually after the year 1965 to about 8%. The interhemispheric transport of air masses is relatively slow and it took two years from the SNAP-9A  $^{238}\text{Pu}$  to reach Finnish Lapland (Miettinen, 1976). The 1986 Chernobyl accident didn't increase the activity ratio in Finnish Lapland. However, in southern Finland the ratio increased to 50%. The  $^{238}\text{Pu}/^{239,240}\text{Pu}$  activity ratio was 5% in top parts and 9% in lower parts of two lichens collected in Inari in 1976 indicating a downward migration of SNAP-9A  $^{238}\text{Pu}$  in lichen thallus (Salminen-Paatero et al., 2014). In the same lichens another Pu activity ratio, namely  $^{241}\text{Pu}/^{239+240}\text{Pu}$ , didn't differ between top and lower parts, the ratio being 7.3–8.5 in all parts. These values for  $^{241}\text{Pu}/^{239+240}\text{Pu}$  ratio are typical for global nuclear weapons testing fallout in 1970's. Third isotope ratio that acts as an indicator of nuclear contamination sources, mass ratio  $^{240}\text{Pu}/^{239}\text{Pu}$ , has been determined from air filters collected in Sodankylä in 1963 and in Rovaniemi in 1965–2011. Mass ratio  $^{240}\text{Pu}/^{239}\text{Pu}$  was 0.14–0.37 in surface air of Sodankylä during the year of deposition maximum from atmospheric nuclear weapons testing (Salminen-Paatero et al., 2012). In the surface air of Rovaniemi, the ratio had the minimum value of 0.117 in 1981 and the maximum value of 0.278 in April–June 1986 (Salminen-Paatero et al., 2019). The mass ratio in air has presumably been the same as in deposition accumulation to lichens of the same area after few years. Most of the  $^{240}\text{Pu}/^{239}\text{Pu}$  mass ratio values in Sodankylä during 1963 and in Rovaniemi 1965–2011 were on the same level as the ratio is in global fallout from nuclear weapons testing.

Biological half-lives of radionuclides are an important factor

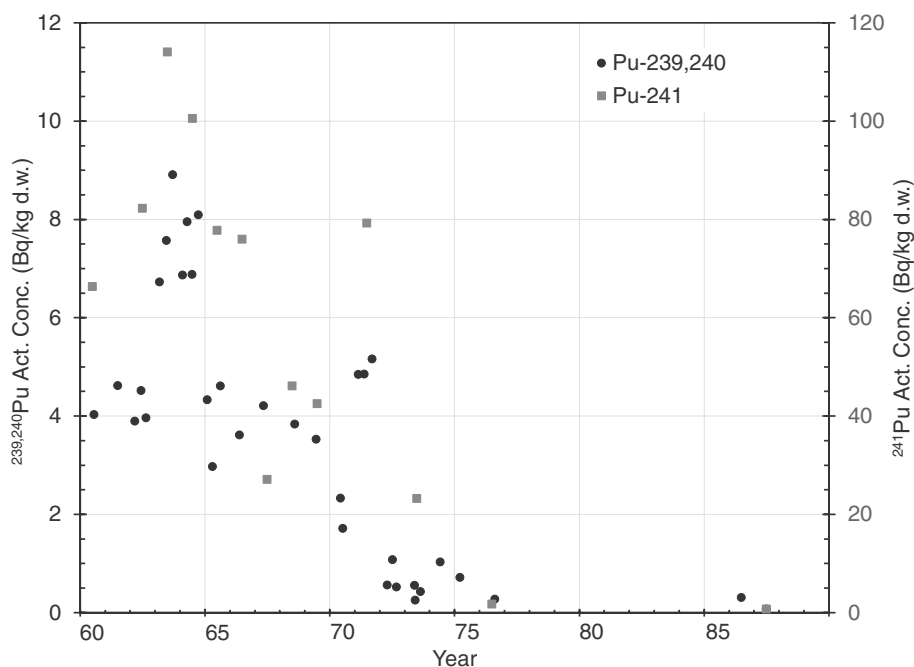


Fig. 3.  $^{239,240}\text{Pu}$  (left vertical axis, black discs) and  $^{241}\text{Pu}$  (right vertical axis, grey squares) in lichen in Finnish Lapland during 1960–1987 (Jaakkola et al., 1981; Paatero et al., 1998).

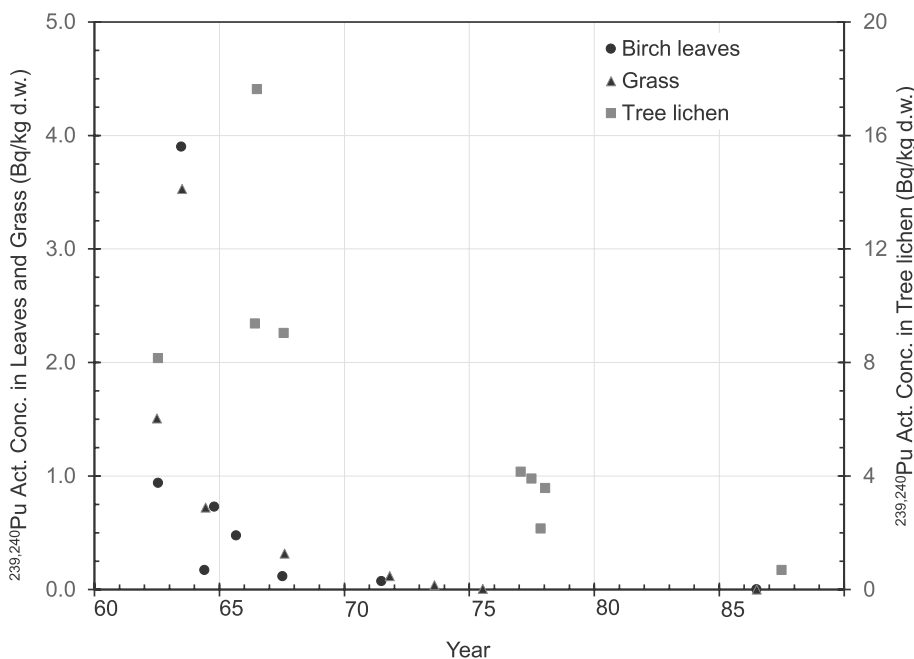


Fig. 4.  $^{239,240}\text{Pu}$  in birch leaves (left vertical axis, black discs), grass (left vertical axis, black triangles) and tree lichen (right vertical axis, grey squares) in Finnish Lapland during 1962–1987 (Jaakkola et al., 1981; Paatero et al., 1998).

describing the long-term behavior of contamination in food chains. In this case it was observed that the biological half-life of plutonium in lichen varied between 480 d and 800 d (Jaakkola et al., 1981). In other words, the amount of plutonium in lichen decreases to half of the original amount in 16–27 months even though its removal by radioactive decay is negligible within the same time scale (the physical half-life of  $^{239}\text{Pu}$  is over 24,000 years). This result is in agreement with the study of Paatero et al. (1998) where a biological half-life value of 730 days was obtained for plutonium in lichen.

The  $^{239,240}\text{Pu}$  activity concentrations in reindeer tissues in Finnish

Lapland between 1963 and 1987 are depicted in Figs. 6 and 7 (Jaakkola et al., 1981; Paatero and Jaakkola, 1998). The highest concentrations are found in liver and the lowest ones in muscle. In 1977, the activity concentrations had decreased to a tenth of the maximum values in 1964–1966, 0.6 Bq/kg fresh weight (f.w.), 0.037 Bq/kg f.w., 0.03 Bq/kg f.w., and 0.007 Bq/kg f.w. in liver, lung, bone, and muscle, respectively. In kidney the  $^{239,240}\text{Pu}$  activity concentrations were similar to those in lung and bone excluding a lone outlier. Due to the low Chernobyl-derived plutonium deposition in Finnish Lapland the  $^{239,240}\text{Pu}$  activity concentration in reindeer liver stayed low after the

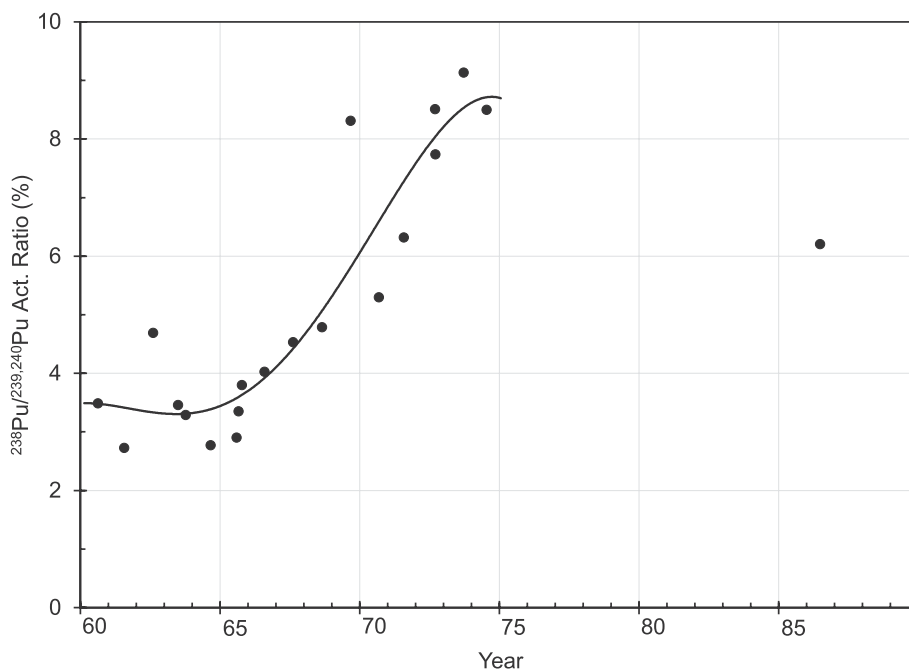


Fig. 5.  $^{238}\text{Pu}/^{239,240}\text{Pu}$  activity ratio in lichen between 1960 and 1986 in Finnish Lapland (Miettinen, 1976; Paatero et al., 1998).

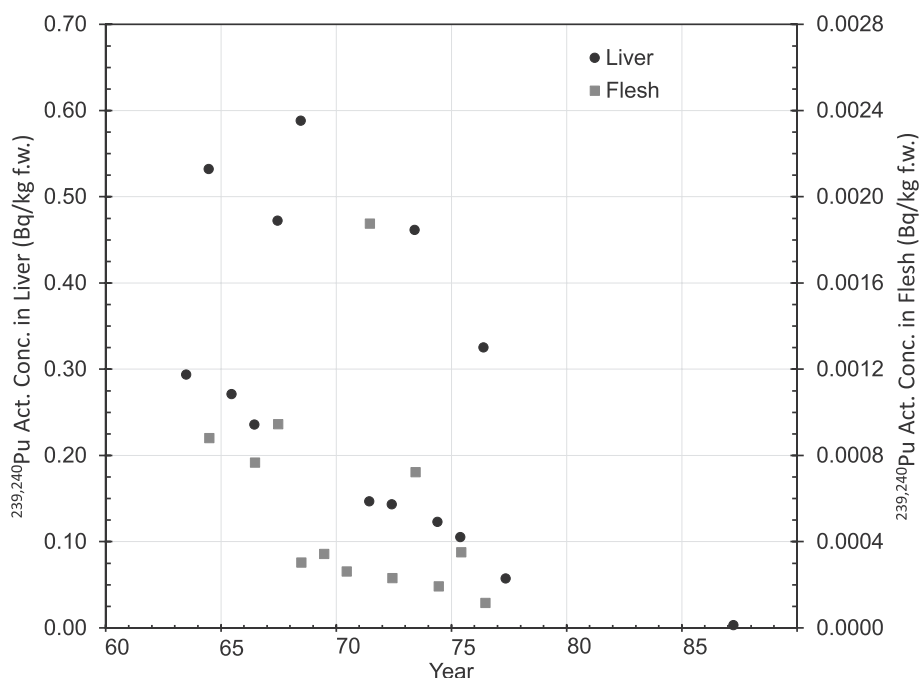


Fig. 6.  $^{239,240}\text{Pu}$  in reindeer liver (left vertical axis, black disks) and flesh (right vertical axis, grey squares) in Finnish Lapland during 1963–1987 (Jaakkola et al., 1981; Paatero and Jaakkola, 1998).

Chernobyl accident. The variations of  $^{239,240}\text{Pu}$  activity concentration in liver of different animals varied by more than an order of magnitude (Jaakkola et al., 1981). The reason for the large variation may be due to differences in lichen and tree lichen consumption of individual animals.

The distribution of plutonium in reindeer was studied by analyzing all major tissues of single reindeers (Fig. 8). In most cases over 60% of the plutonium was in the liver, and, thus, the plutonium content of reindeer depends mainly on the amount of plutonium in the liver. The plutonium content of bones varied between 10% and 40% from one case to another (Hakanen, 1981; Jaakkola et al., 1981).

A comparison of plutonium content of different tissues in reindeer

and elk revealed that in 1964–1966 the plutonium contents of lungs were about the same (Keinonen et al., 1977; Miettinen et al., 1980). However, in liver and bone the  $^{239,240}\text{Pu}$  concentration was approximately an order of magnitude higher in reindeer than in elk. This is explained by the fact that elks don't consume lichen as the reindeer do. Reindeer is the only large animal known to have received the major portion of its plutonium body burden from a dietary source instead of inhalation. The fraction of plutonium absorbed from the reindeer fodder along the gastrointestinal (GI) tract has been estimated to be  $(4.0 \pm 0.4) \times 10^{-5}$  (Jaakkola et al., 1981). In a later study of Paatero and Jaakkola (1998) a value of  $3.0 \times 10^{-5}$  was obtained.

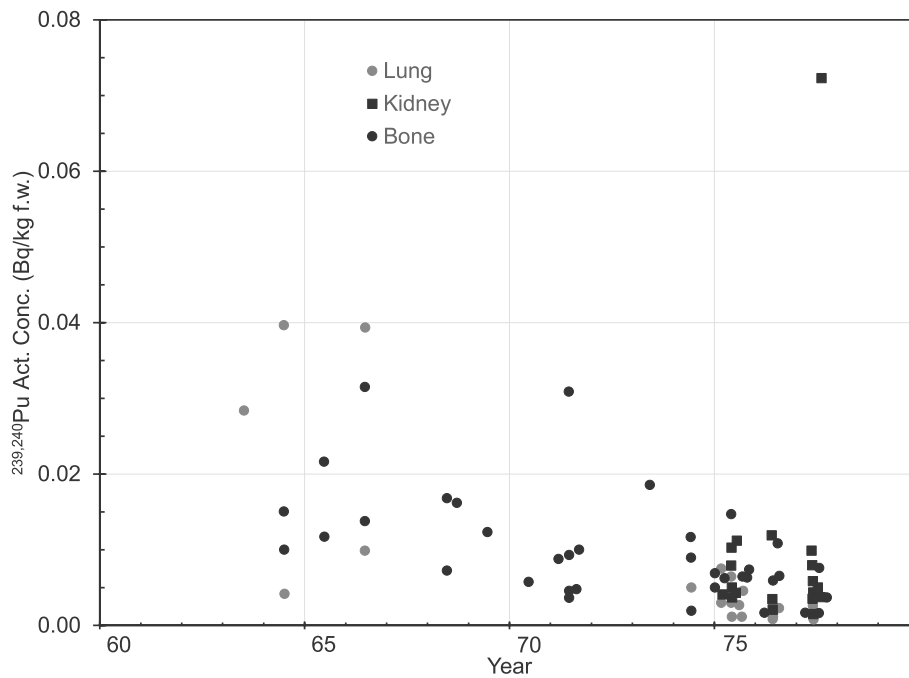


Fig. 7. <sup>239,240</sup>Pu in reindeer tissues (lung: grey disks, kidney: black squares, bone: black disks) in Finnish Lapland during 1963–1977 (Jaakkola et al., 1981).

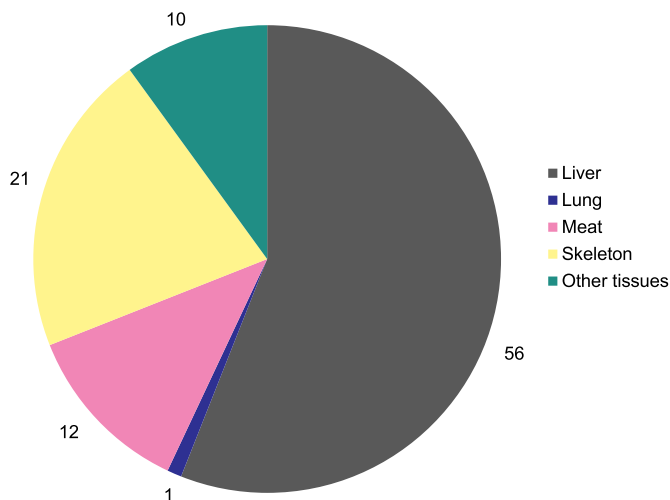


Fig. 8. Distribution of plutonium between different reindeer tissues (%). This particular animal was a 13 years old male that was slaughtered in 1974 (Jaakkola et al., 1981).

The intake of plutonium by Sami people in Finland was assessed to be 1.5 Bq/year for men and 0.59 Bq/year for women in 1967 due to reindeer tissue consumption. In 1975 the corresponding figures were 0.37 Bq/year and 0.22 Bq/year (Jaakkola et al., 1981). About 70–80% of the plutonium intake was due to the consumption of reindeer liver. For comparison, the annual plutonium intake of residents in the New York City area, USA, was estimated to be 0.055 Bq/year in 1972. Plutonium content of humans was studied in the works of Mussalo-Rauhamaa (1981) and Mussalo-Rauhamaa et al. (1984). Autopsy samples of five Samis were dissected at the Central Hospital of Rovaniemi during 1977–1979. Plutonium content of the samples was analysed using radiochemical separation methods followed by semiconductor alpha spectrometry. The <sup>239,240</sup>Pu activity concentration in lung, liver and bone samples are depicted in Fig. 9. The highest activity concentrations are found in liver. There seems to be no clear relationship between the <sup>239,240</sup>Pu activity concentration and the profession of the person.

A significant portion (37–68%) of the total amount of <sup>239,240</sup>Pu in human body is in liver (Fig. 10). The Sami whole body content doesn't differ from that of southern Finns. This is an interesting feature because the diet of Sami contains 10–15 times more plutonium than the diet of southern Finns. The fractional absorption factor of <sup>239,240</sup>Pu from the human gastrointestinal tract was calculated to be  $8 \times 10^{-4}$ .

#### 4. Americium

Americium-241 contents of lichen samples collected in Finland (Fig. 11) are about a quarter of the corresponding <sup>239,240</sup>Pu contents. The <sup>241</sup>Am/<sup>239,240</sup>Pu ratios are gradually increasing because of the <sup>241</sup>Pu decaying to <sup>241</sup>Am, being 0.04–0.32 in Finnish lichens during time period 1962–1974 (Harva, 1976). In the Chernobyl deposition this ratio will be 2.81 at its maximum in 2058 (Paatero et al., 1994). The biological half-life of <sup>241</sup>Am, 320 days (Paatero et al., 1998) is shorter than that of plutonium. The <sup>241</sup>Am/<sup>239,240</sup>Pu activity ratio in reindeer liver, 0.18, is close to the values found in lichen (Jaakkola et al., 1981). However, in reindeer bones the ratio was 1.0 indicating a more efficient accumulation of americium into bones compared to the accumulation of plutonium. The same applies to reindeer lung (Paatero and Jaakkola, 1998). A GI absorption coefficient of  $7.5 \times 10^{-4}$  for reindeer was observed. This indicates that americium is transferred from fodder to reindeer almost an order of magnitude more efficiently than plutonium. All these data indicate, that americium is more mobile in the biosphere compared with plutonium.

#### 5. Curium

Only negligible amounts of curium isotopes were produced in the nuclear tests (Holm and Persson, 1978b). After the Chernobyl accident  $0.55 \pm 0.08$  Bq/kg d.w. of <sup>242</sup>Cm was found in lichen collected in Inari (Paatero, 2000). This was less than one per cent of the values observed in southwestern Finland. Even this small amount of <sup>242</sup>Cm was rather quickly removed from the lichen-reindeer-man food chain due to the short half-life of <sup>242</sup>Cm, 163 days. Based on the average <sup>243,244</sup>Cm/<sup>242</sup>Cm activity ratio of 0.0056 in lichen in southern Finland and the observed <sup>242</sup>Cm activity concentration one can calculate that the <sup>243,244</sup>Cm activity concentration in lichen in northern Finland was about 0.003 Bq/kg d.w.

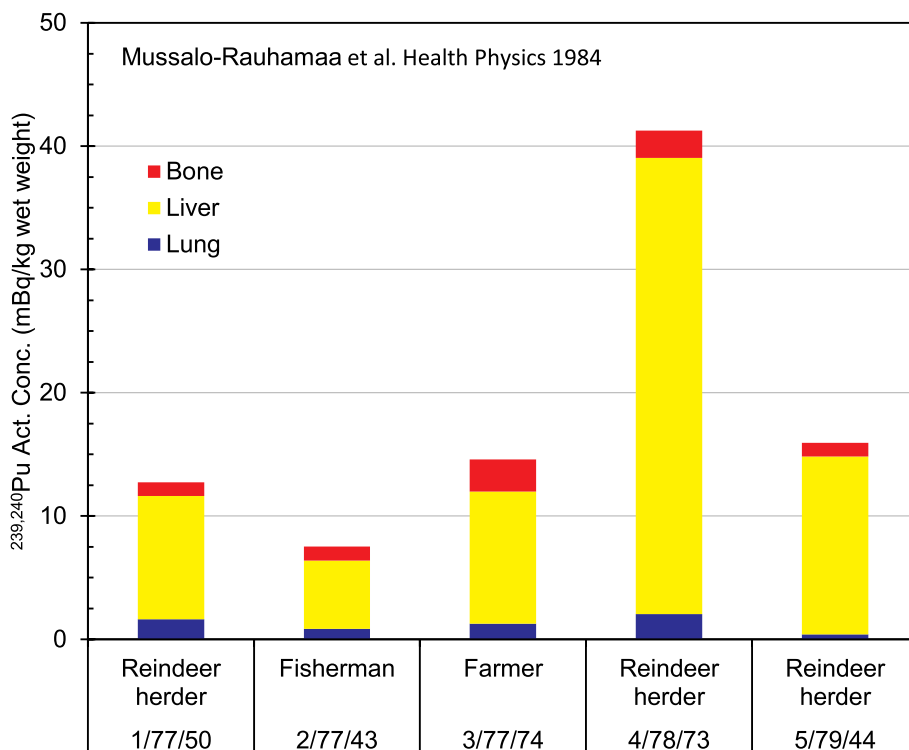


Fig. 9.  $^{239,240}\text{Pu}$  activity concentration in lung, liver and bone of five Sami persons (Mussalo-Rauhamaa et al., 1984). The sample codes beneath the profession: Sample No./Year of death/Age.

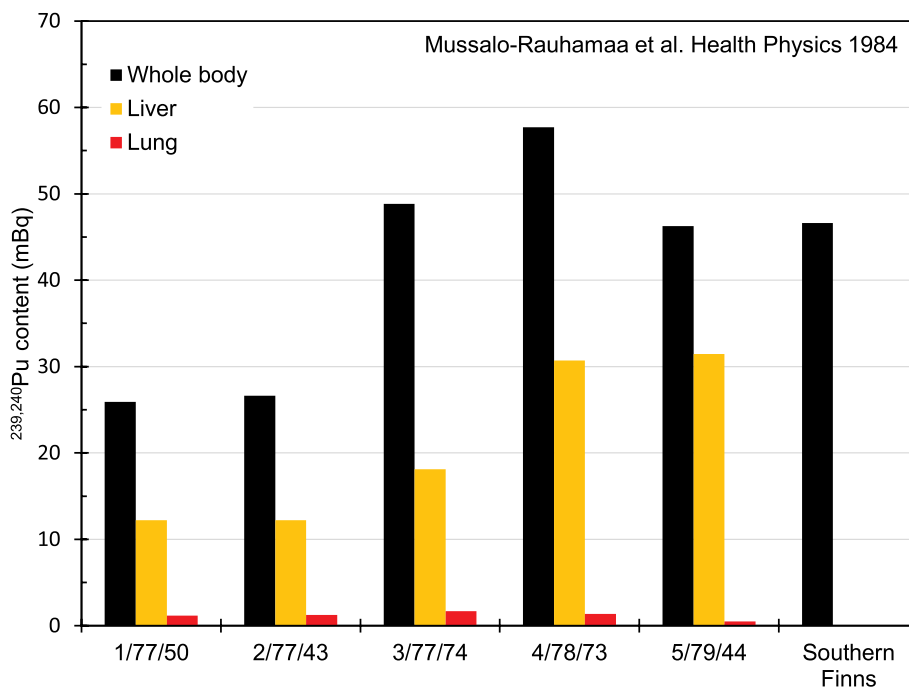


Fig. 10.  $^{239,240}\text{Pu}$  activity content of lung, liver and whole body of five Sami persons and average whole body content of southern Finns (Mussalo-Rauhamaa et al., 1984). The sample codes: Sample No./Year of death/Age.

(Paatero et al., 1998). The  $^{242}\text{Cm}$  content of a grass sample collected in summer 1986 at Inari region, northern Finland was  $0.11 \pm 0.03 \text{ Bq/kg d.w.}$  but the  $^{242}\text{Cm}$  content of birch leaves remained under the detection limit of  $0.10 \text{ Bq/kg d.w.}$  Due to the chemical similarity of curium and americium one can expect that the transfer of curium along the lichen-reindeer-man food chain is similar to that of americium.

### 6. Neptunium

Neptunium-237 is an extremely long-lived radionuclide with a half-life of 2.14 million years. Due to analytical difficulties, especially the lack of suitable tracers, it has gained very little attention within studies of environmental radioactivity. The only publication of environmental



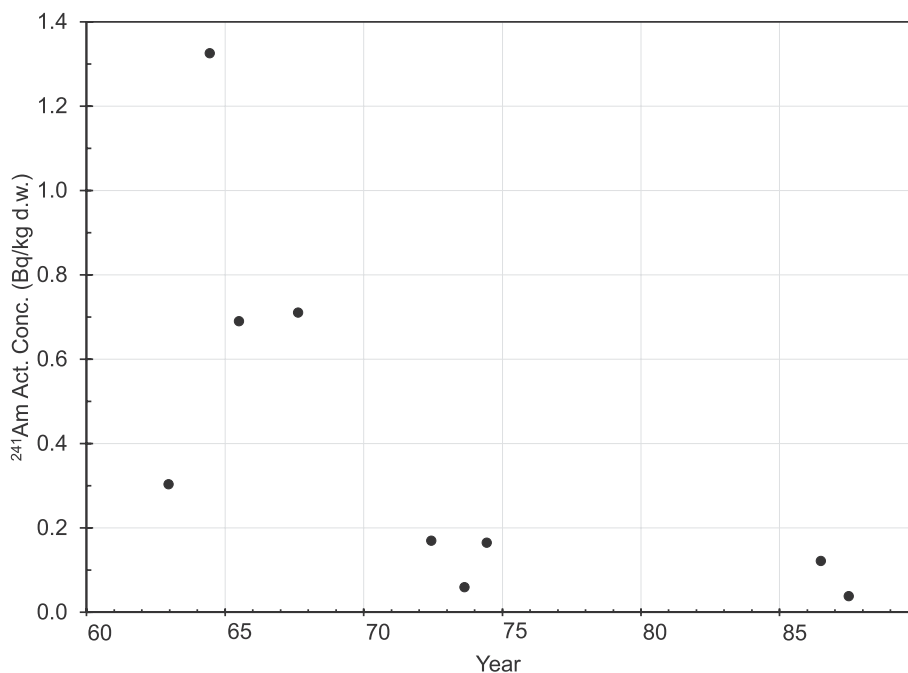


Fig. 11. <sup>241</sup>Am in lichen in Finnish Lapland during 1962–1987 (Jaakkola et al., 1981; Paatero et al., 1998).

<sup>237</sup>Np in Finland is that of Salminen et al. (2009). In Rovaniemi and Inari the <sup>237</sup>Np content of lichen was below detection limit, <0.76 mBq/kg d. w. Lindahl et al. (2004) analysed two lichen samples collected in northern Sweden close to the Finnish-Swedish border. The <sup>237</sup>Np activity contents of lichen samples from 1984 to 1987 were  $10.9 \pm 0.9$  mBq/kg and  $1.60 \pm 0.13$  mBq/kg, respectively. Based on the observed <sup>238</sup>Pu/<sup>239,240</sup>Pu activity ratio of  $0.04 \pm 0.01$  in the 1987 sample it is evident that it contained practically no contamination from the Chernobyl accident.

In addition, a short-lived isotope <sup>239</sup>Np ( $t_{1/2}$  2.4 d) was detected in mushroom (240 Bq/kg f.w.) and wild leafy vegetables (58 Bq/kg f.w.) in southern Finland after the Chernobyl accident (Rantavaara, 1987).

## 7. Conclusions

Considerable amounts of data concerning the transfer of transuranium elements along the subarctic food chain lichen-reindeer-man have been gathered during the last five decades. Globally this food chain is a rare one as there are very few other food chains that enrich radioactivity from deposition to man with a similar efficiency. But still, the amount of transuranium elements in humans is quite low compared to fission products and natural radionuclides. The lichen-reindeer-man food chain has been quite unique because it has allowed studies of the transfer of radionuclides into man in an actual environment without added tracers and test animals.

Luckily the Chernobyl accident affected significantly only the southernmost reindeer herding district of Halla. The accident timing effects on the deposition pattern has been studied with dispersion calculations. If the accident had happened only one day earlier than it actually did the resulting <sup>137</sup>Cs deposition in the Sami region would have been 20–50 kBq/m<sup>2</sup>, ten times more than the global weapons test fallout (Siljamo and Lahtinen, 2006). This would have ruined the reindeer husbandry and the associated culture for decades.

Despite the additional radionuclide intake of Sami population due to reindeer meat consumption the incidence rate of common cancers in Finland (e.g. prostate, breast and skin) has been traditionally less than within the Finland's main population (Soininen and Pukkala, 2016). This difference seems to fade away gradually. One reason may be related to the lifestyles. When the radioactivity studies began six decades ago

many reindeer herding Samis were still living in a subsistence economy based on local reindeer meat, fish, game, berries etc. But slowly their nutrition habits have changed towards the main population's diet.

In the current situation the concentrations of transuranium elements decrease constantly in the environment, excluding <sup>241</sup>Am and <sup>237</sup>Np, the daughter and granddaughter nuclide of short-lived beta emitter <sup>241</sup>Pu ( $t_{1/2}$  14.35 a), which is abundant particularly from Chernobyl and Fukushima (nuclear power plant) accidents. In general, the environmental concentrations of transuranium radionuclides approach in many cases detection limits of the instruments. Natural radionuclides pose higher risk in respect to radiation dose and exposure in environment and along food chains, compared to the artificial transuranics. However, future situation is unknown. For maintaining emergency preparedness and effective radiation protection of vulnerable Arctic ecosystems, well-established, reliable and sensitive detection methods - both radio-analytical and instrumental - will be always needed and need to be improved for monitoring transuranium elements in the environment.

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## Appendix A. Supplementary data

Supplementary data to this article can be found online at <https://doi.org/10.1016/j.jenvrad.2019.106126>.

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